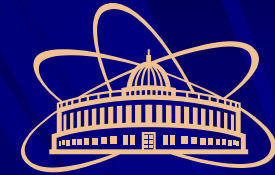


*Flerov Laboratory of Nuclear Reactions
Joint Institute for Nuclear research*



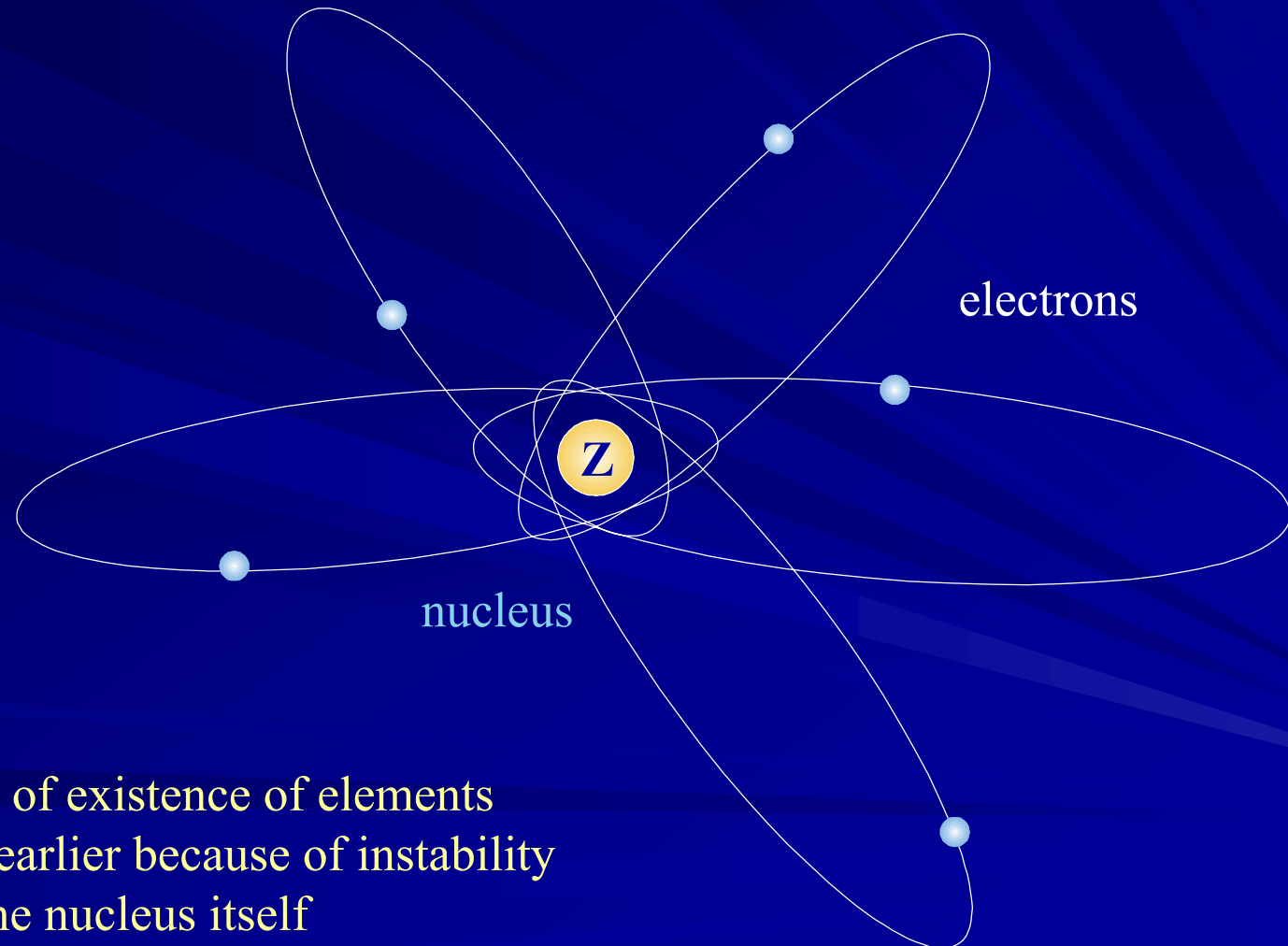
Mikhail Itkis

Superheavy Elements
current status and future trends

Frontiers in the Physics of Nucleus, Peterhof 2005

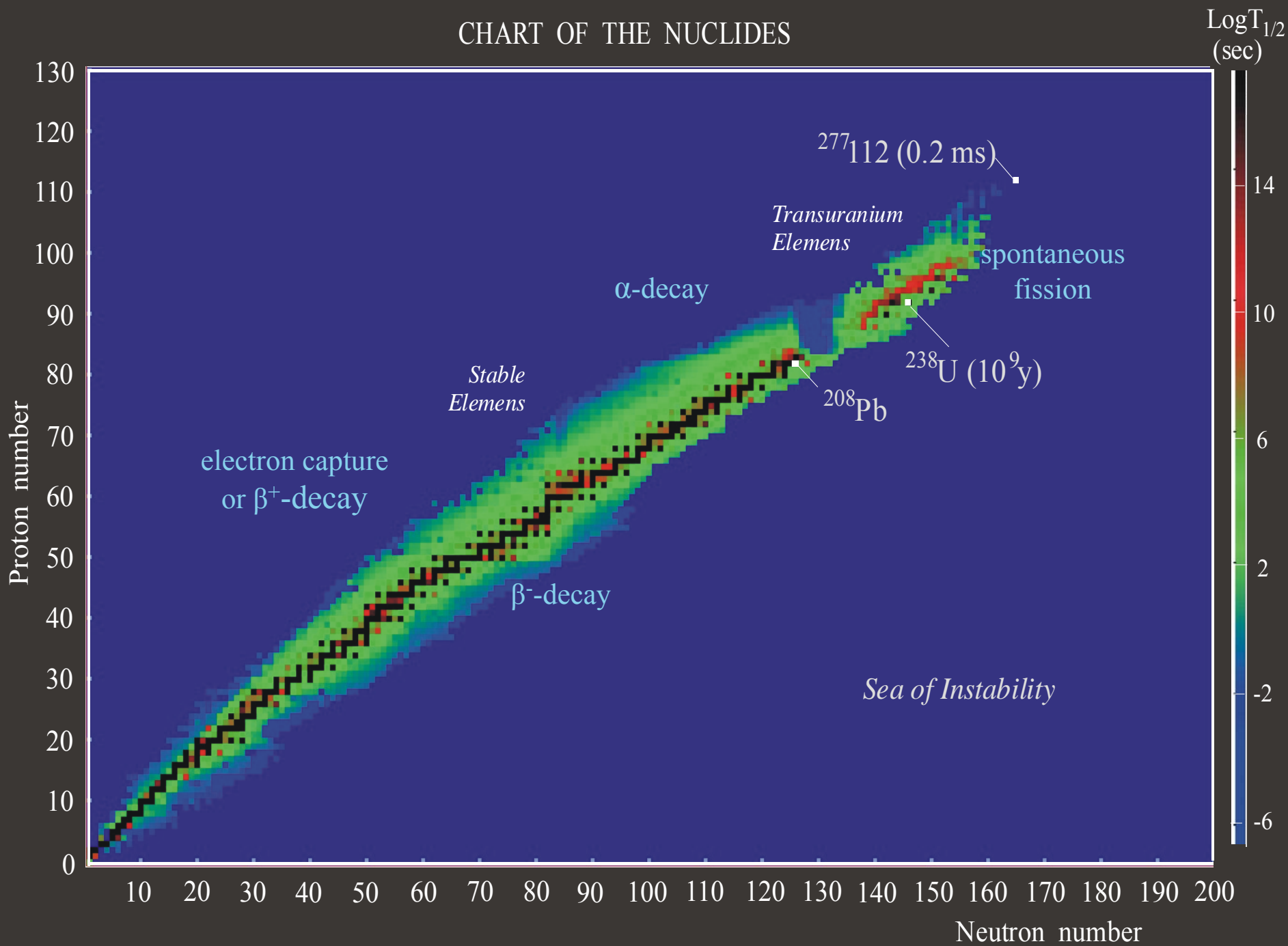
E. Rutherford (1932)

According to QED such an atomic structure is valid for very heavy atoms with $Z \sim 170$ or even more



...but the limit of existence of elements is reached much earlier because of instability of the nucleus itself

CHART OF THE NUCLIDES



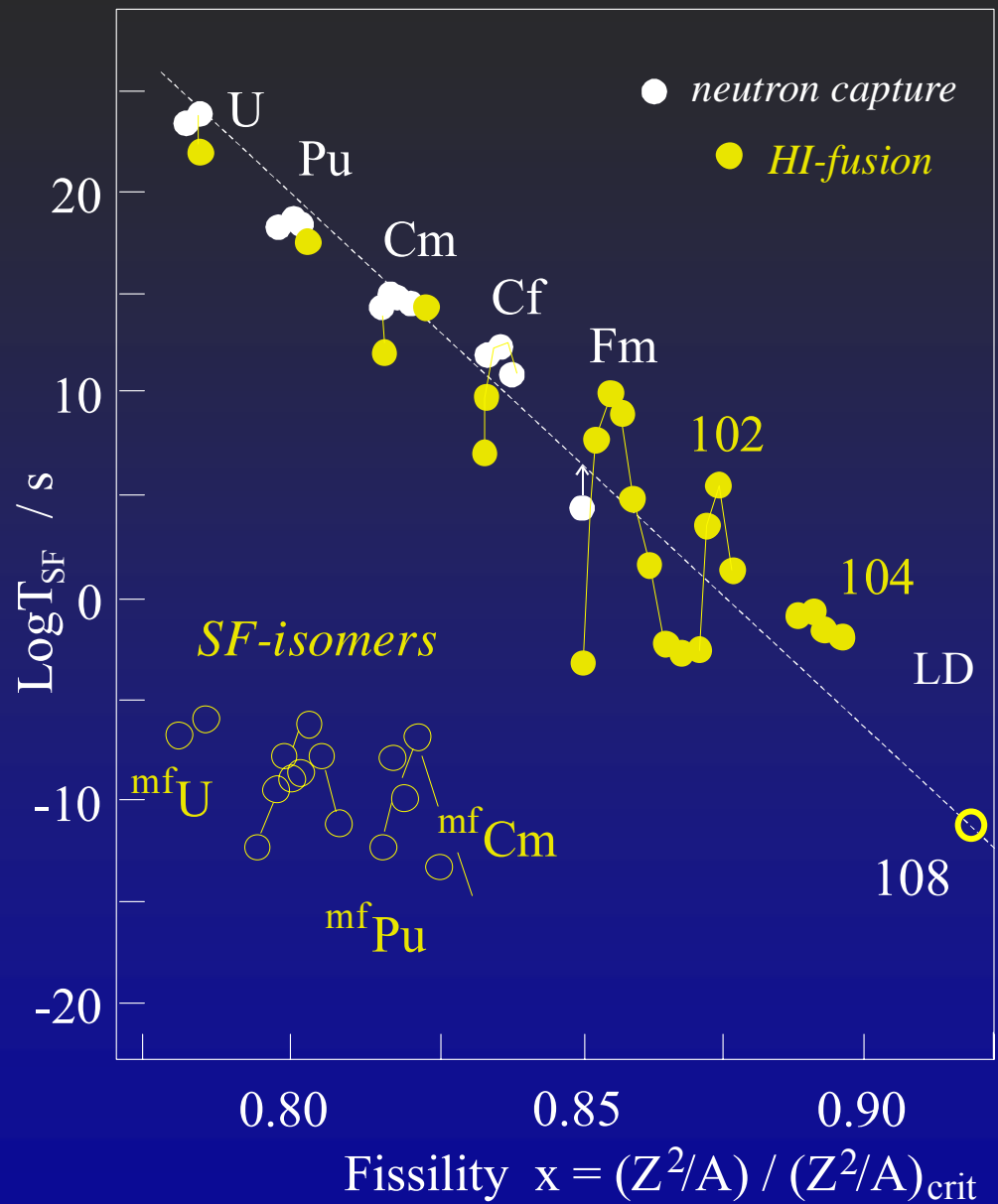
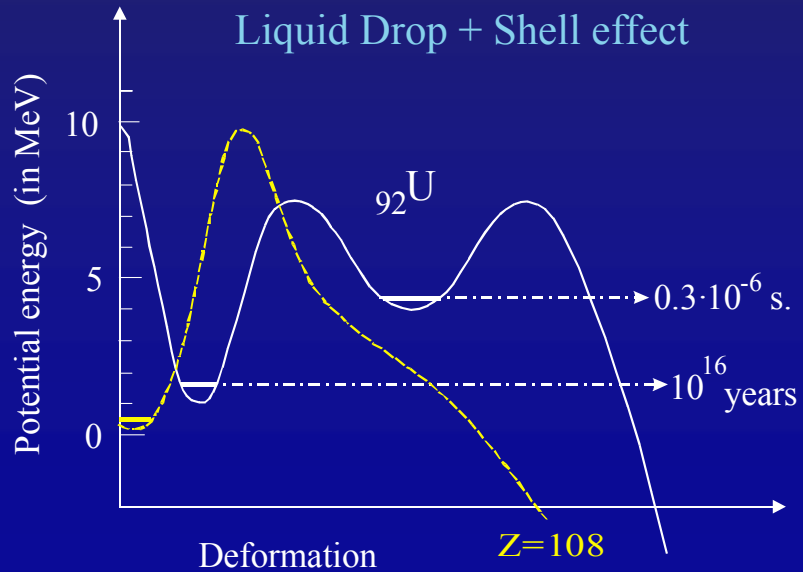
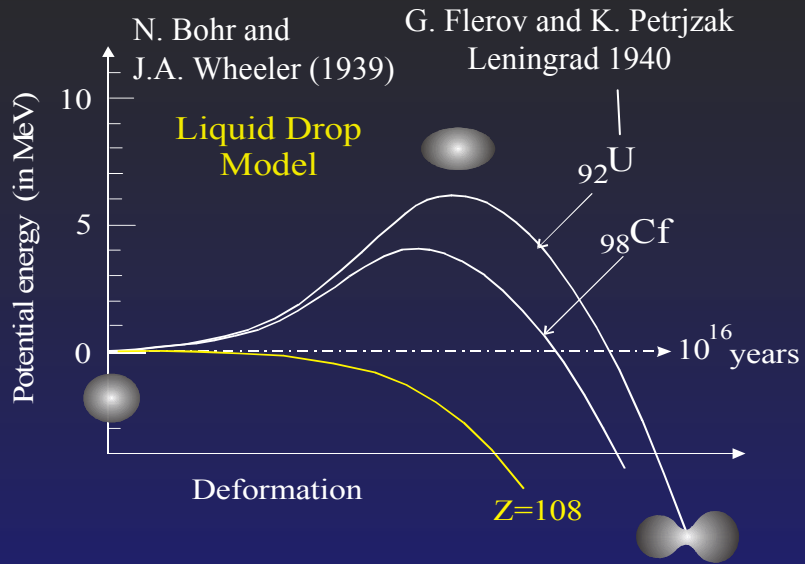


Chart of shell corrections to the LD-potential energy

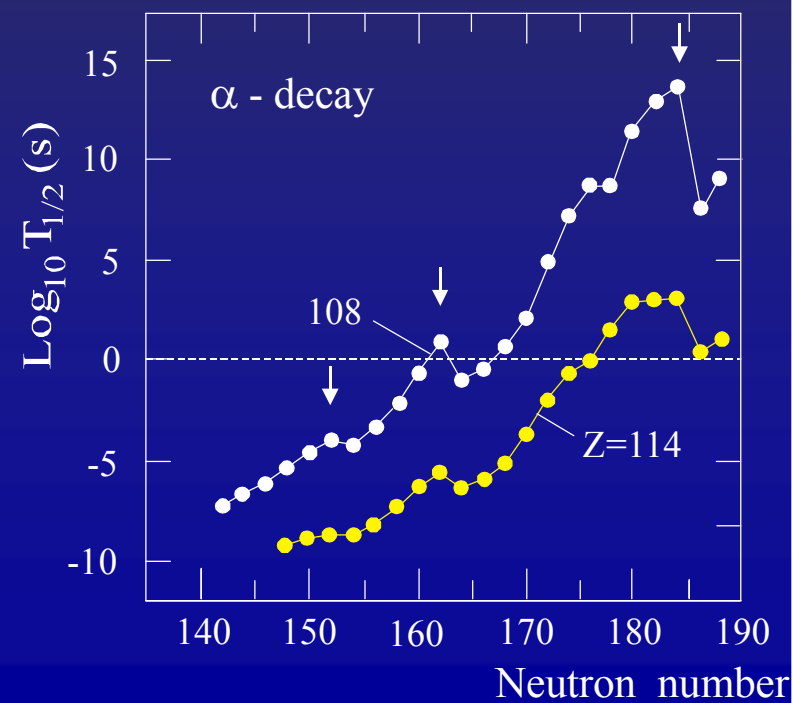
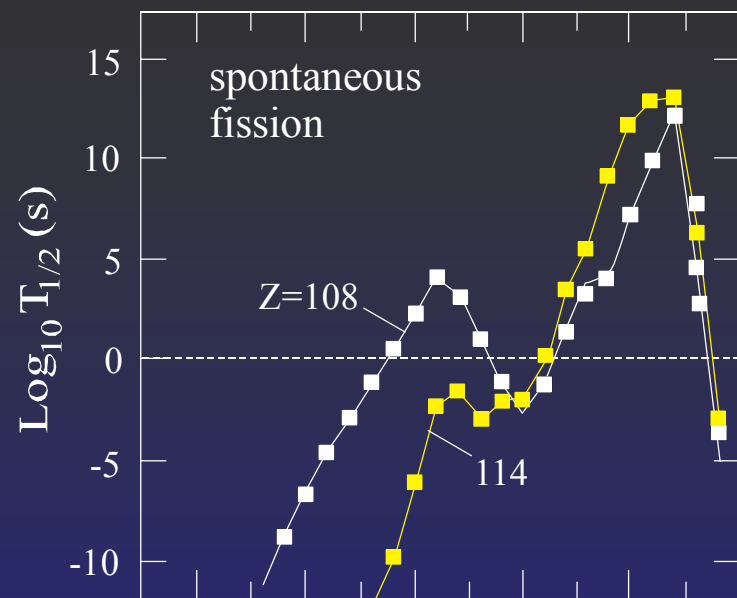
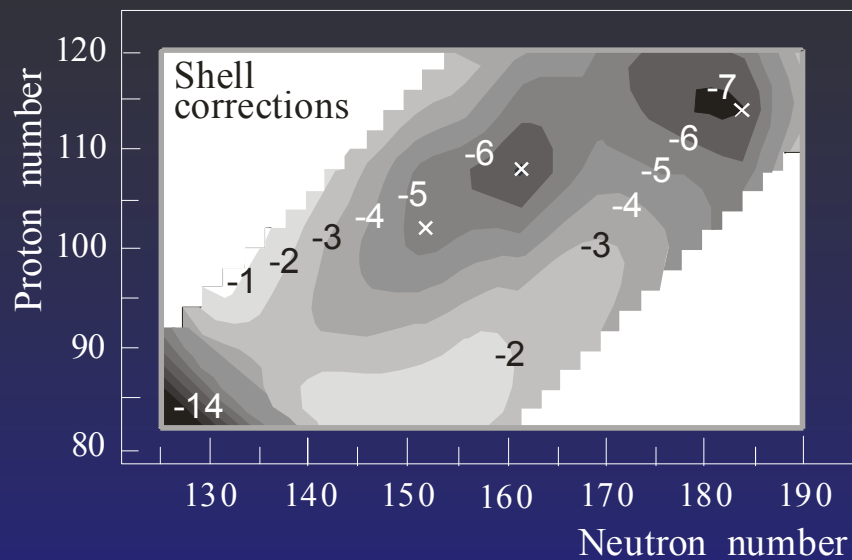
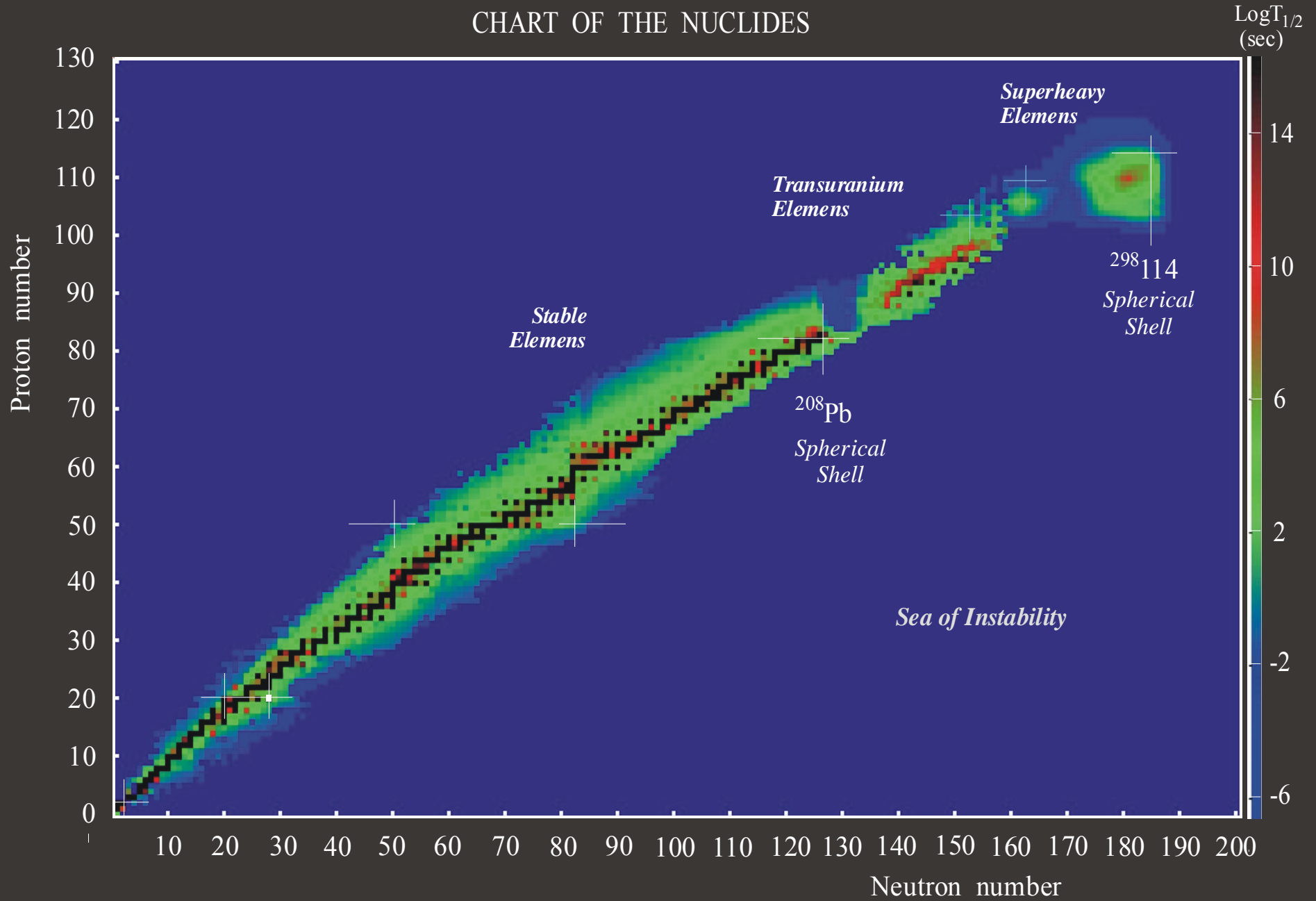


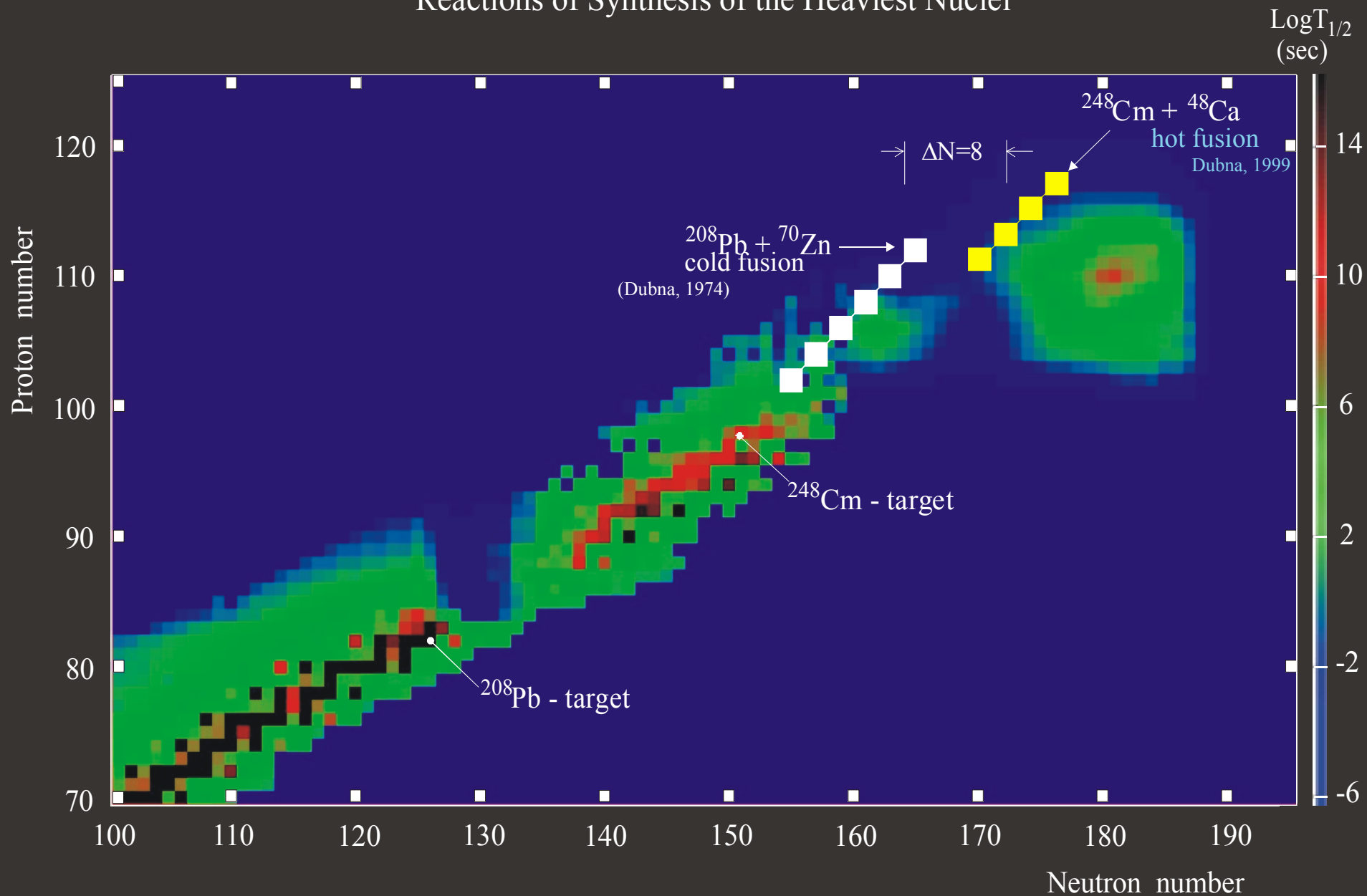
CHART OF THE NUCLIDES



How to check the idea about the existence of hypothetical “Islands of Stability” in the region of yet unknown super heavy elements?

- reactions of the SHE synthesis *key problem*
- what are we expecting to see in the experiment *unusual properties of SHE*
- what we have already observed *decay modes of SHE*
- setting the experiments *synthesis of elements 113,115 and 118*
- Chemistry of SHE *identification of atomic numbers of SHE*
- overall picture of SHE *nuclear shells and stability of the SHE*
- the search for surviving SHE. Prospects

Reactions of Synthesis of the Heaviest Nuclei



Here there are two questions:

What is the fusion probability for ^{48}Ca and actinide nuclei?

What is the survival probability of the compound nucleus with $Z=114-118$ at the excitation energy $E^* \geq 30$ MeV?

Fusion probability

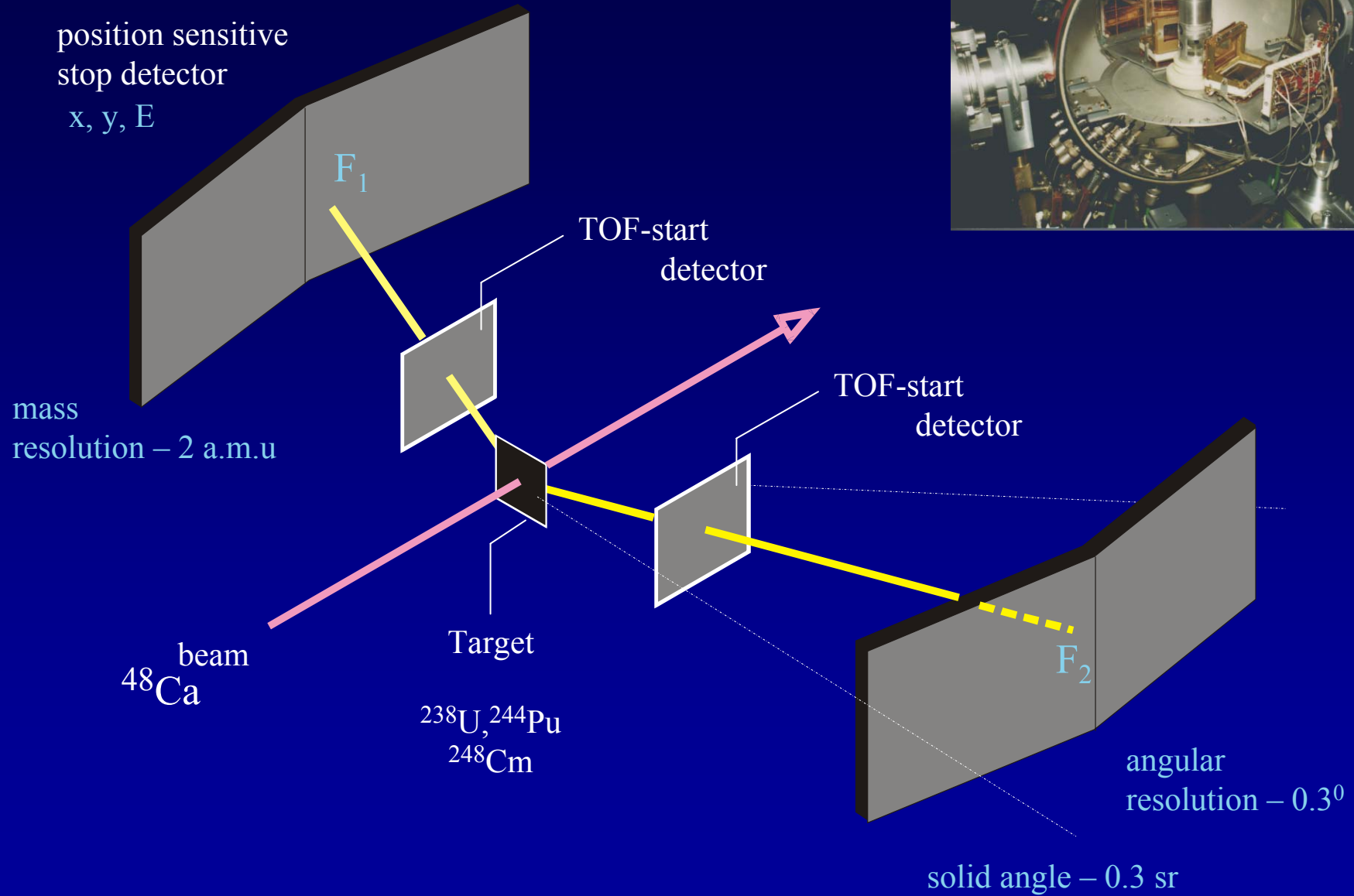
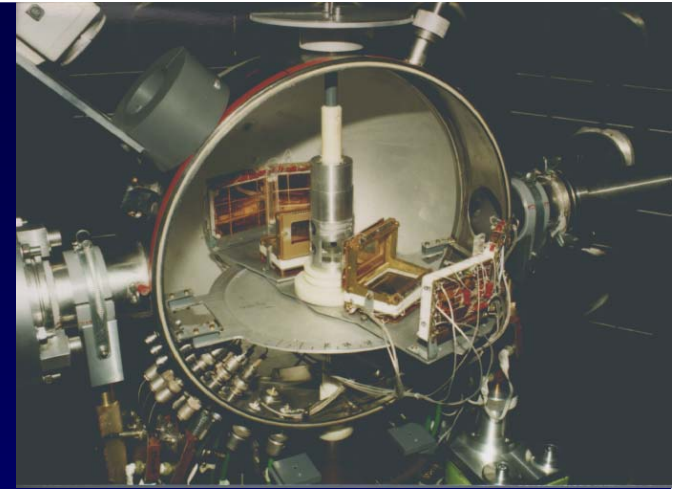
Let us consider the fusion of the ^{48}Ca and ^{248}Cm occurred and resulted in the formation of the compound nucleus $^{296}116$ with an excitation energy of about 40 MeV

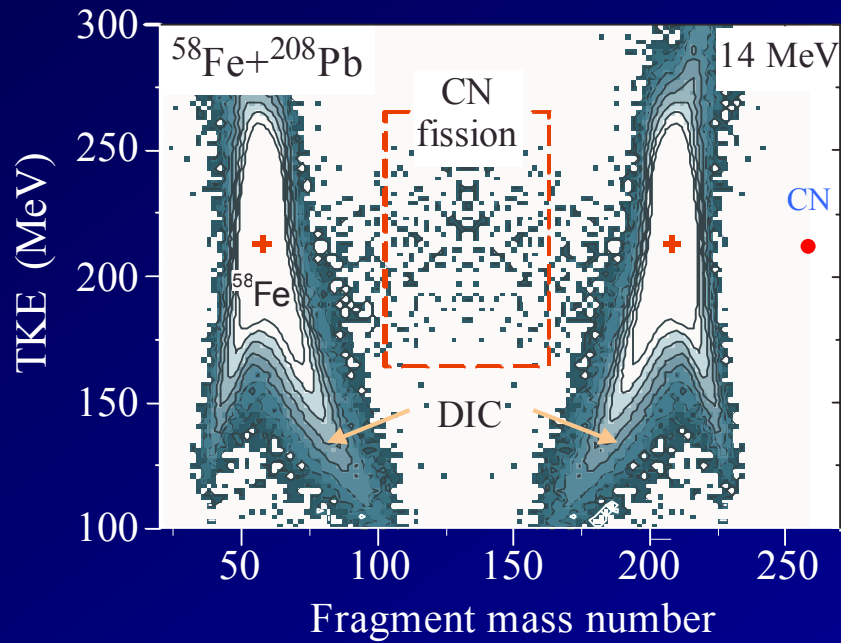
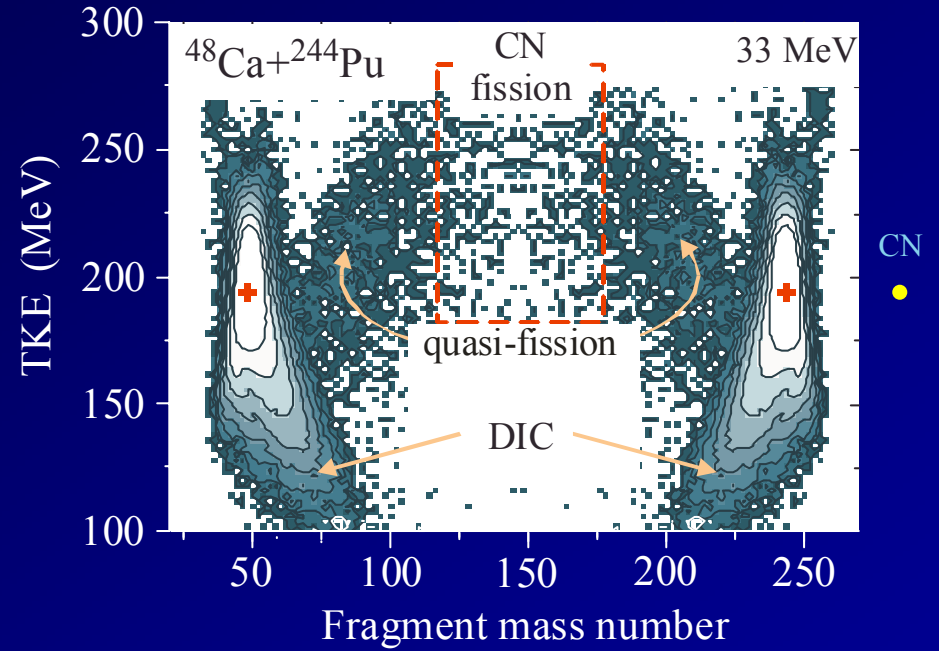
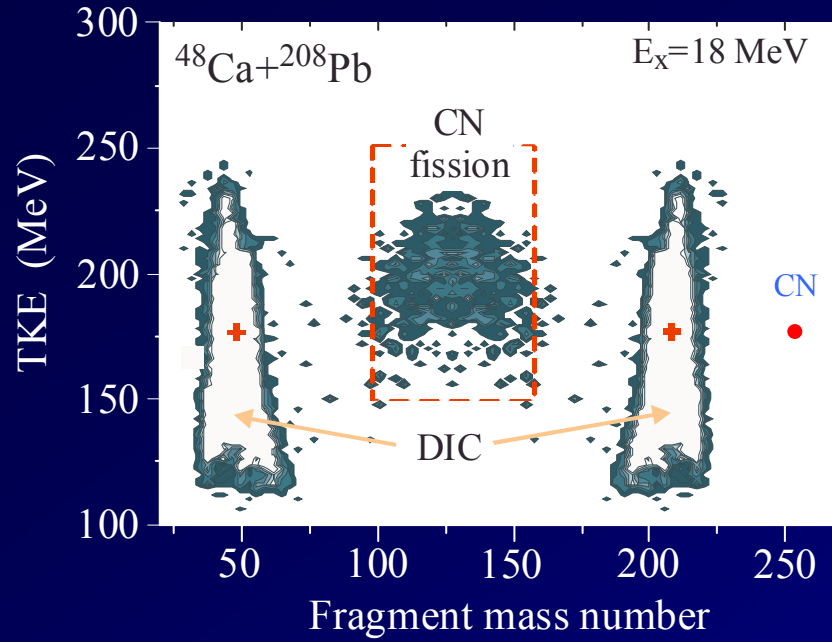
Evidently, the dominant decay mode of such a nucleus would be fission into two fragments

Accordingly, one could attempt investigating the probability of formation of the compound nucleus by measuring its fission characteristics.

In other words, one should measure mass and energy distributions of the fission fragments in the kinematics that corresponds to the full momentum transfer from the ^{48}Ca -projectile to the composite system with $A=296$

CORSET setup

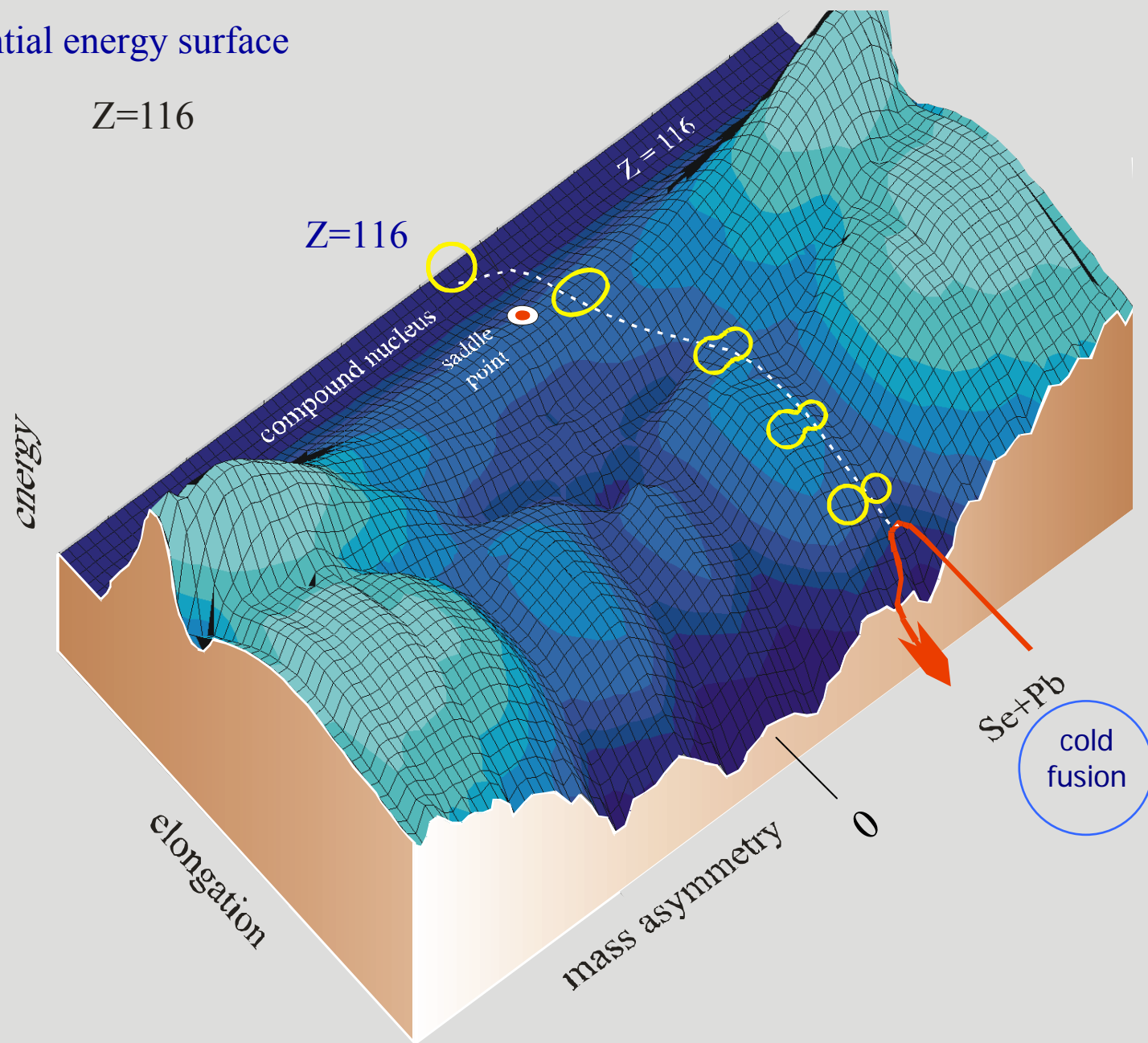




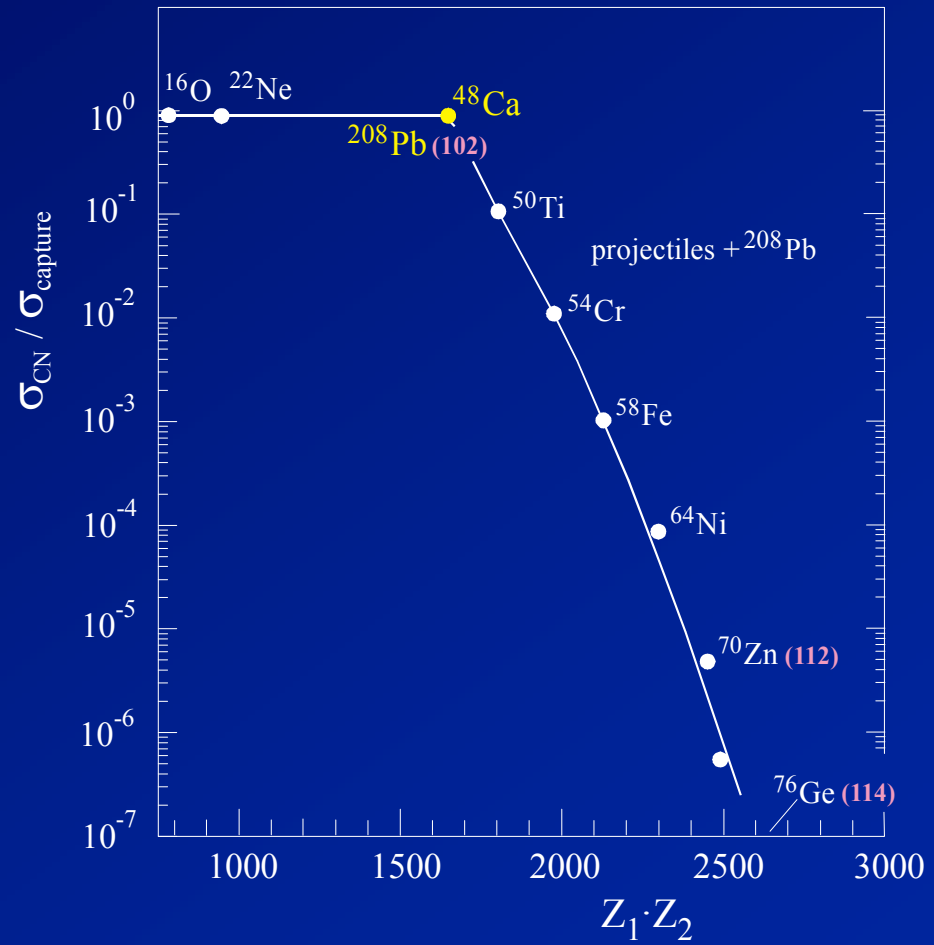
M. Itkis et al., (2002)
Fragment Energy and Mass Distributions
in Cold and Hot Fusion Reactions

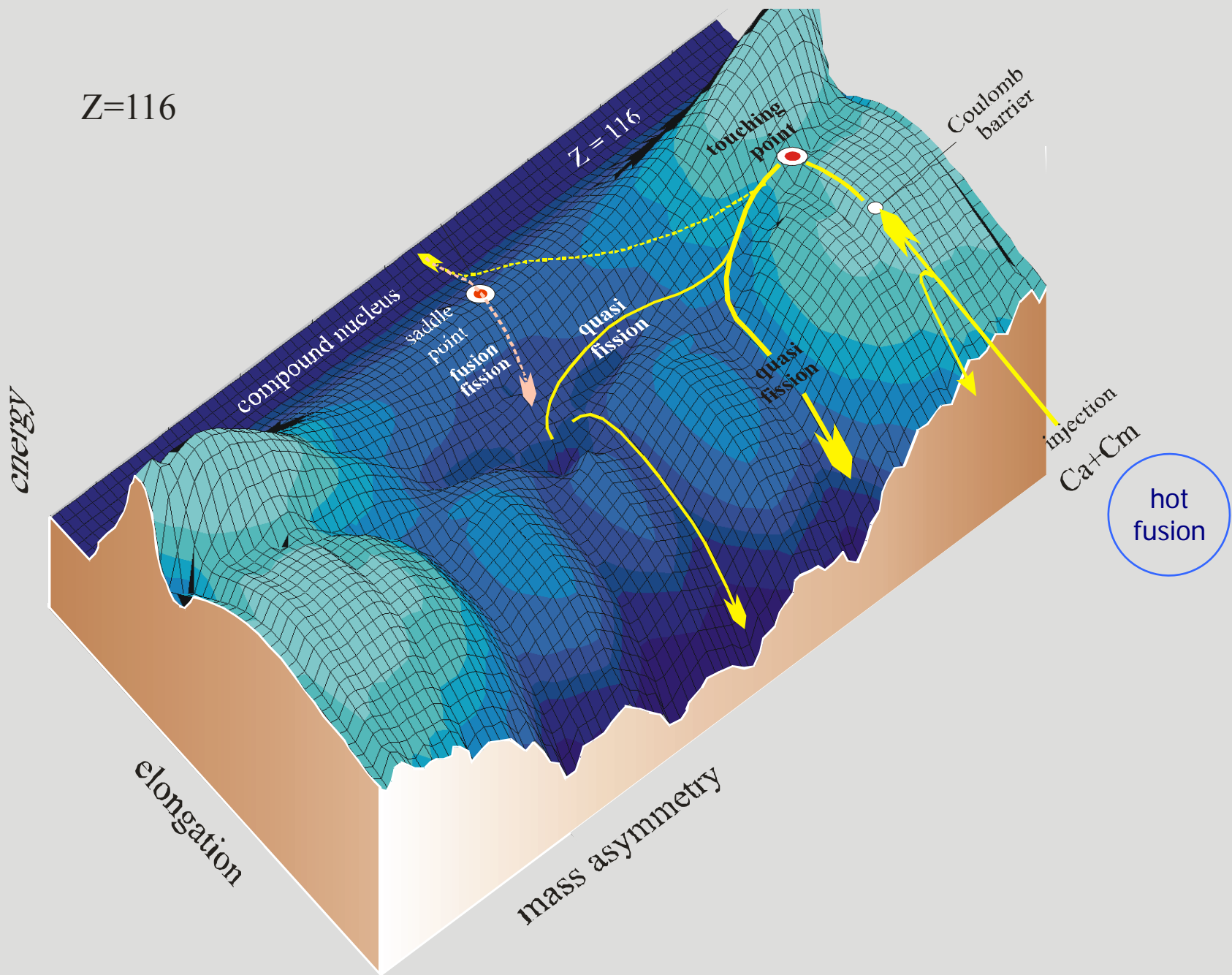
Potential energy surface

Z=116

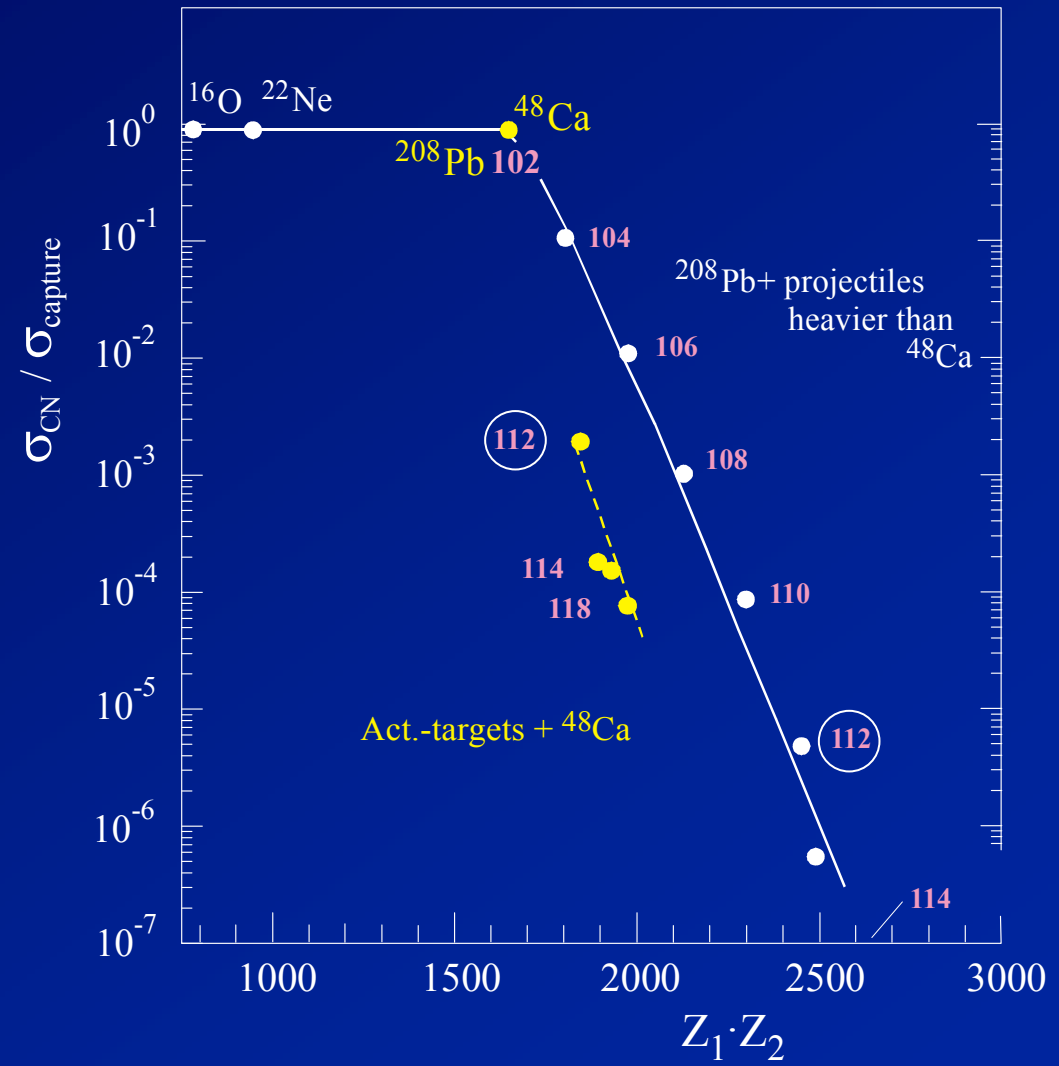


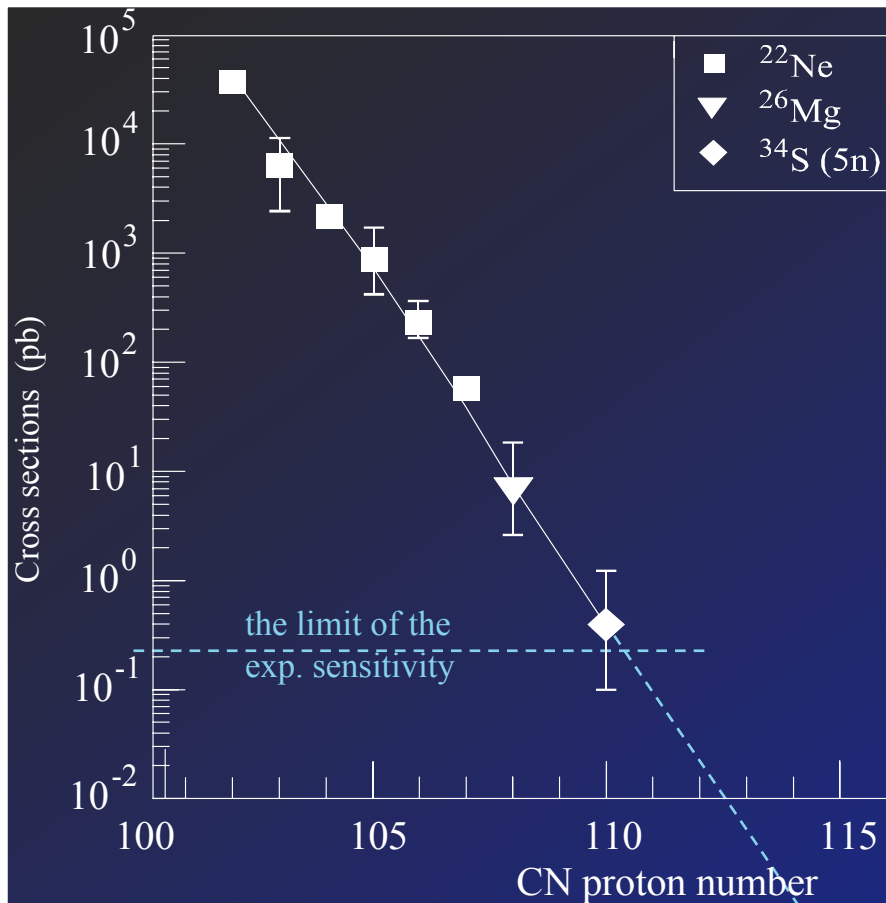
Fusion probability





Fusion probability



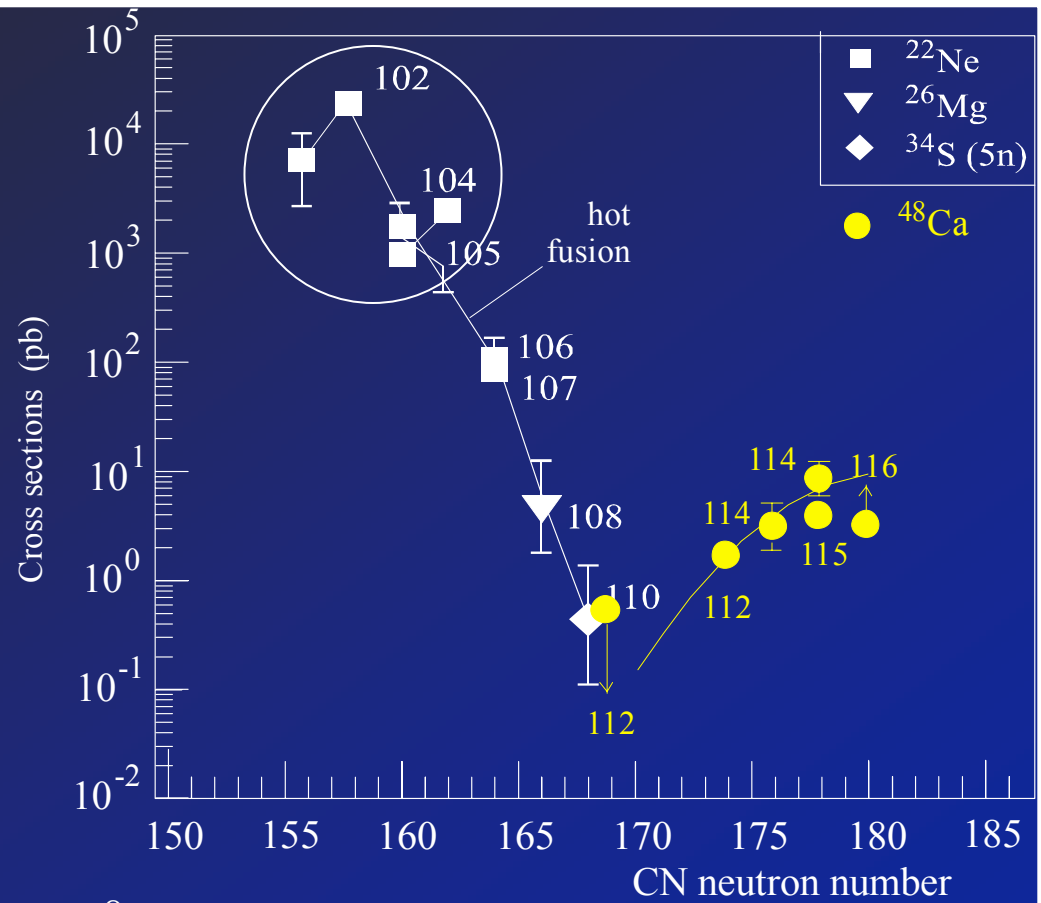


Survival probability

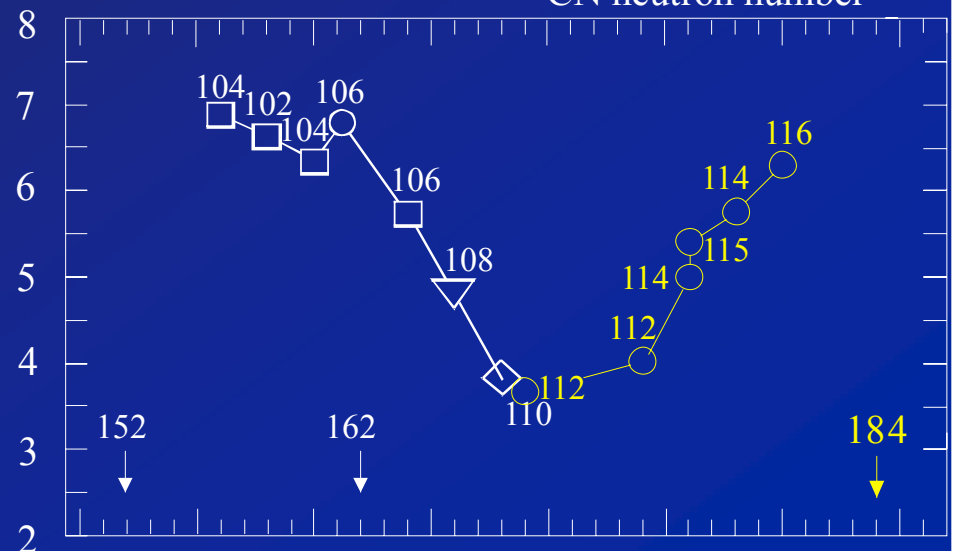
$$\sigma_{\text{xn}} = P_{\text{xn}} \cdot \prod_{i=1}^{i=x} (\Gamma_n / \Gamma_f)_i \sim (\Gamma_n / \Gamma_f)^x$$

$$(\Gamma_n / \Gamma_f)_i \sim \exp[(B_f - B_n) / T]$$

where $B_f = \underbrace{B_f^{\text{LD}}}_{0} + \Delta E^{\text{Shell}}$



Fission barriers (MeV)



Higher neutron number in the compound nucleus increases its survivability considerably

The survivability of the compound nucleus, i.e., the $\sigma_{\text{EVR}}(\text{N})$ is an independent evidence for the stabilizing effect of the **N=184 shell in the domain of SHE**

Natural occurrence of Ca isotopes (in %):

$^{40}\text{Ca} - 96.94$

$^{42}\text{Ca} - 0.647$

$^{43}\text{Ca} - 0.135$

$^{44}\text{Ca} - 2.086$

$^{46}\text{Ca} - 0.004$

$^{48}\text{Ca} - 0.187$

x 400

Consumption
of ^{48}Ca (68%) - 0.5 mg/h

$\rightarrow \text{Ca}^{5+}$

beam
intensity - $4 - 8 \cdot 10^{12}/\text{s}$

beam time - 4000 h/y

Isotopes:
 $\text{U}[233, 238], \text{Pu}[242, 244], \text{Am}[243], \text{Cm}[245, 247], \text{Cf}[249] + ^{48}\text{Ca} \rightarrow Z = 112 - 118$

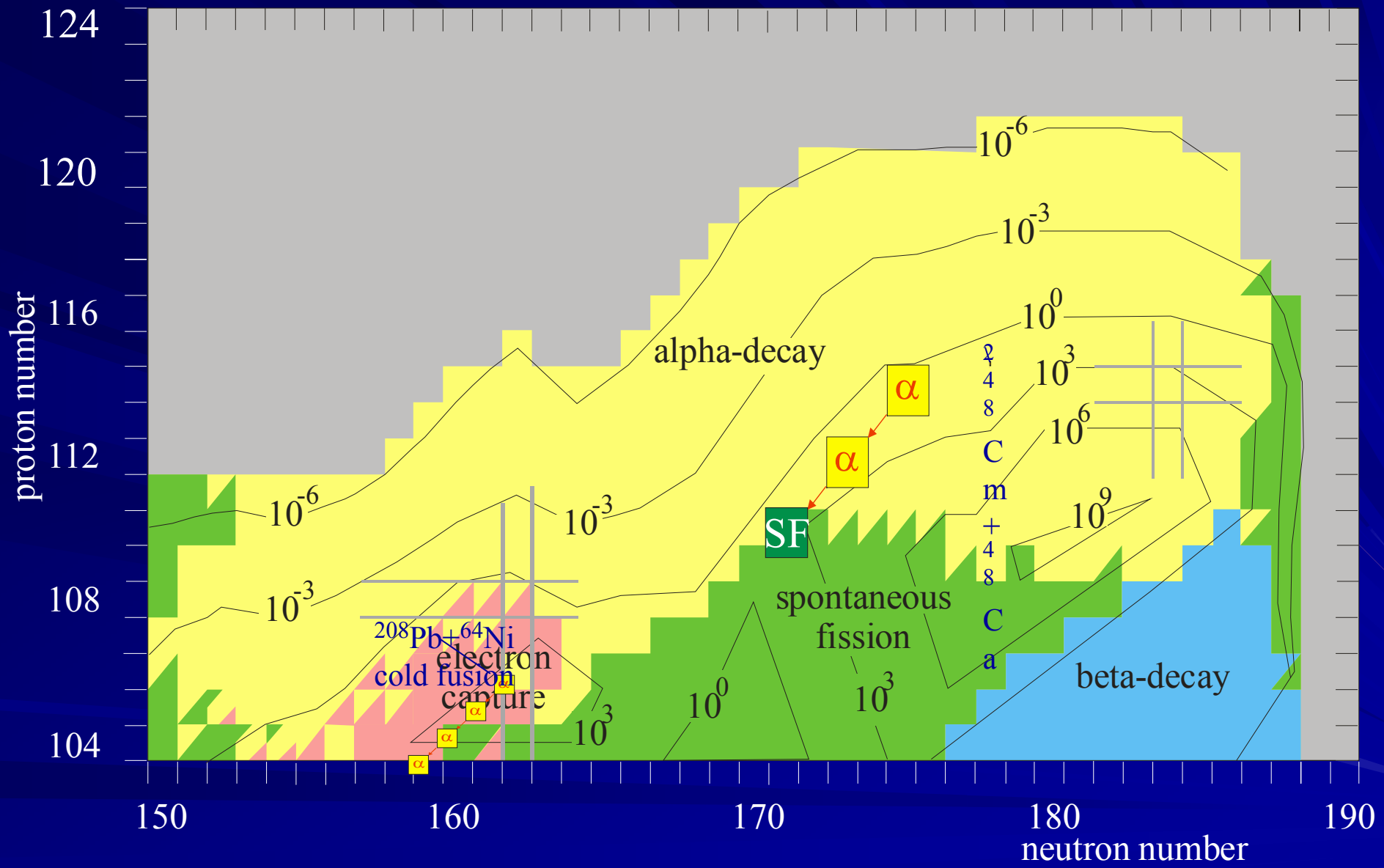
technology of the target
preparation - $0.3 \text{ mg}/\text{cm}^2$

Separation of super heavy nuclei and
detection of their radioactive decays

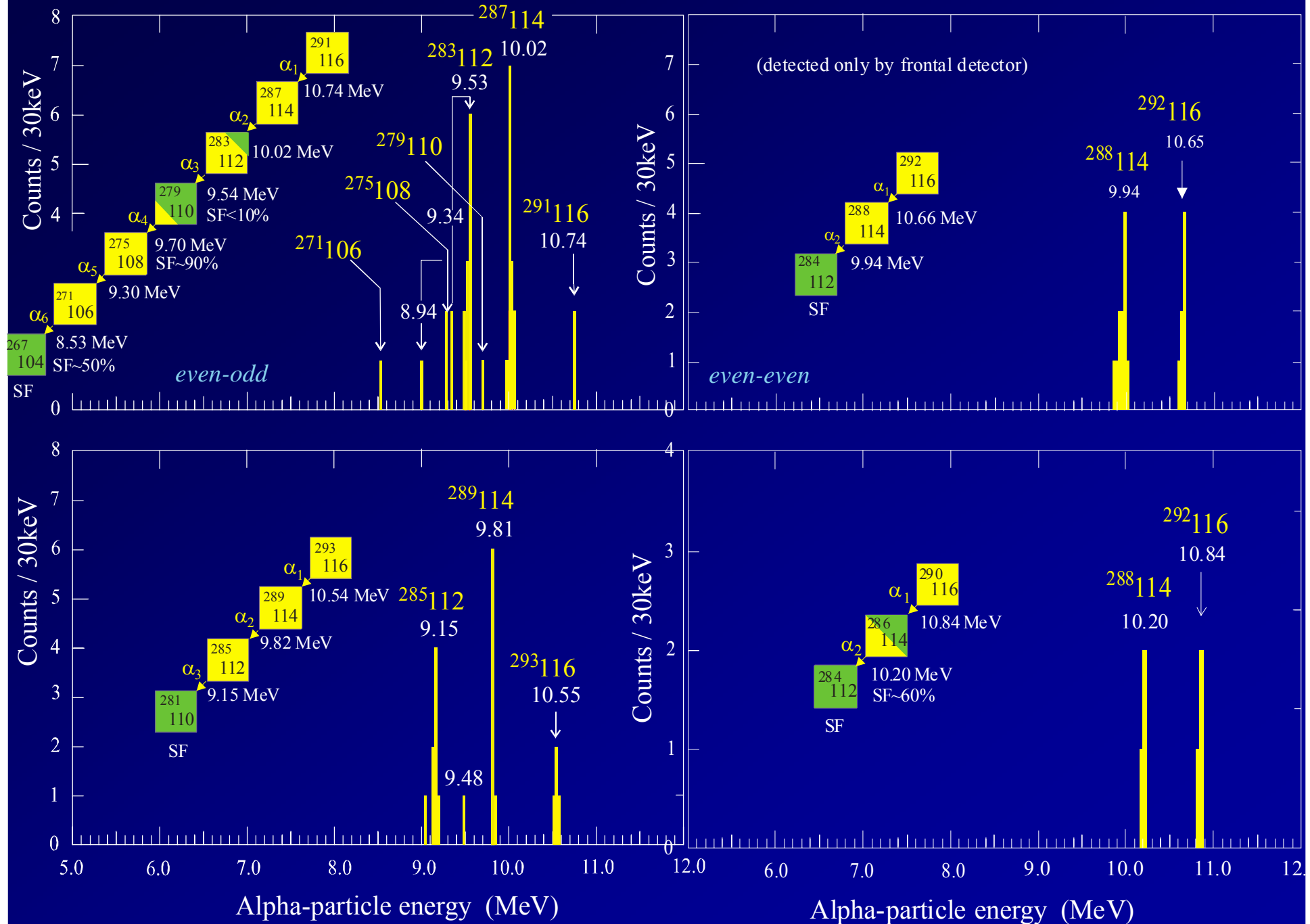
isotope enrichment 98-99%
S-2 separator
(Sarov)

now: DGFRS

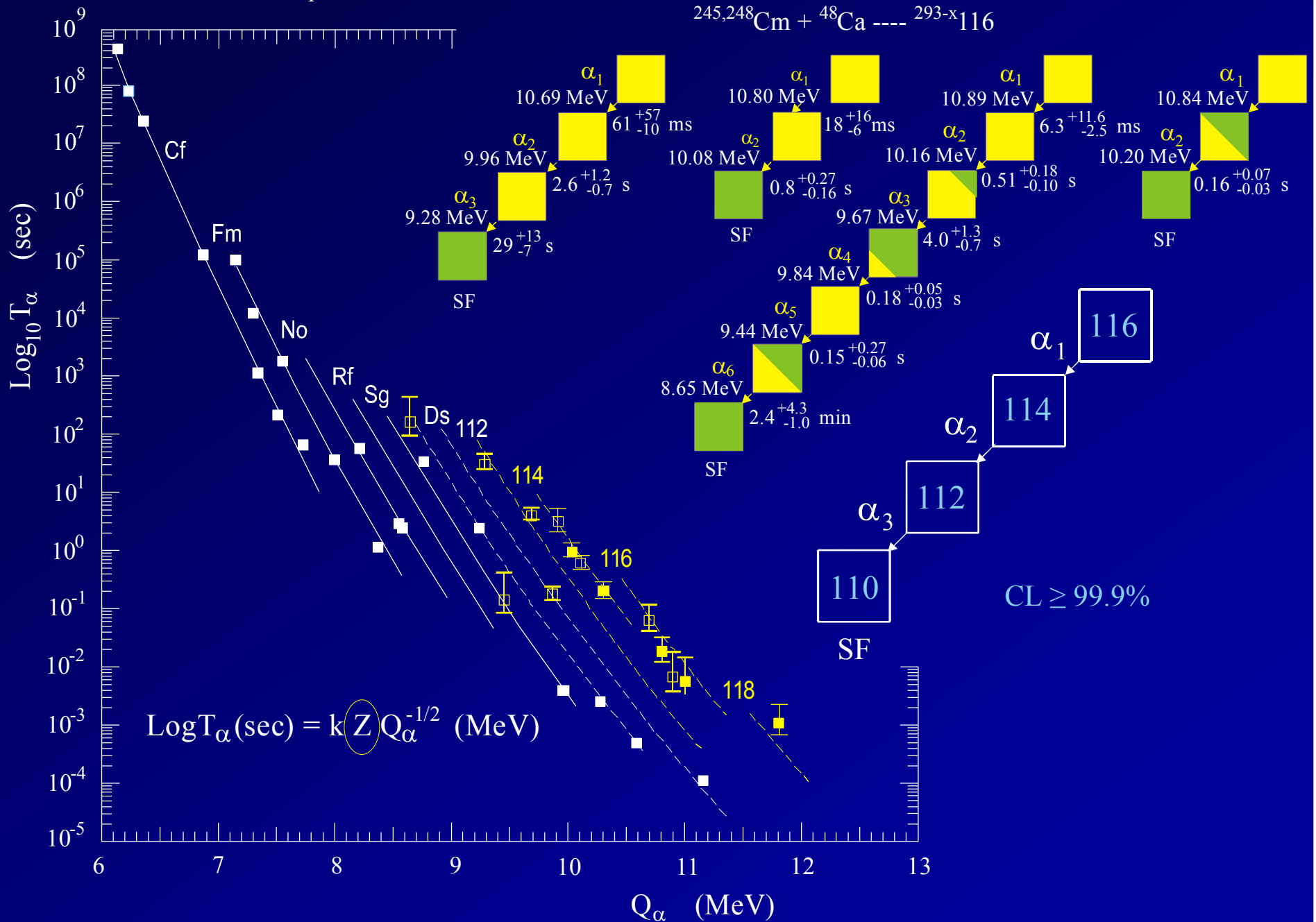
isotope production
high flux reactors
(Oak Ridge, Dimitrovgrad)

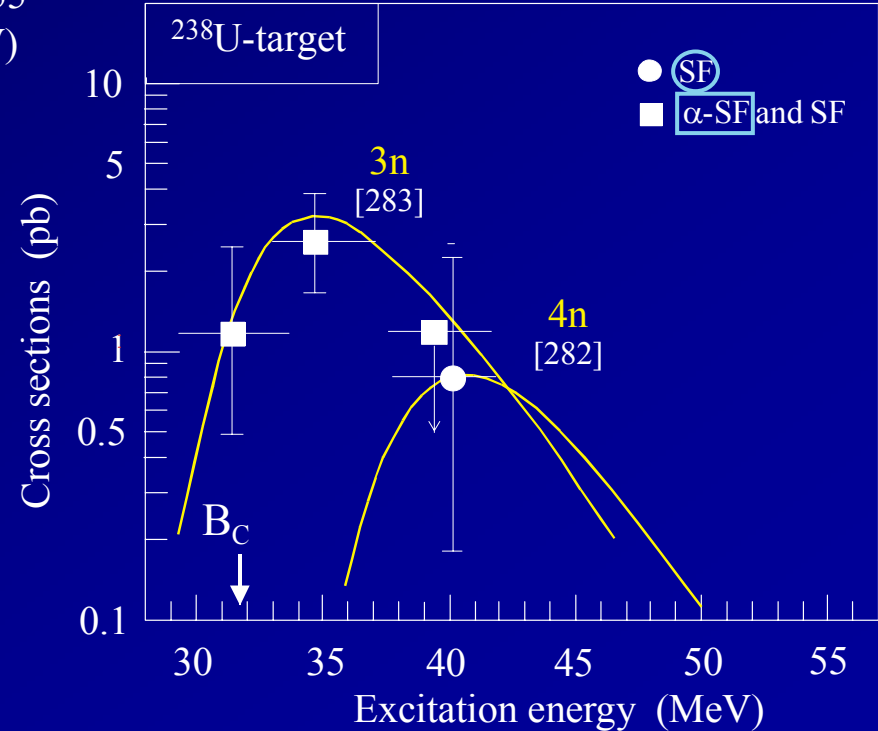
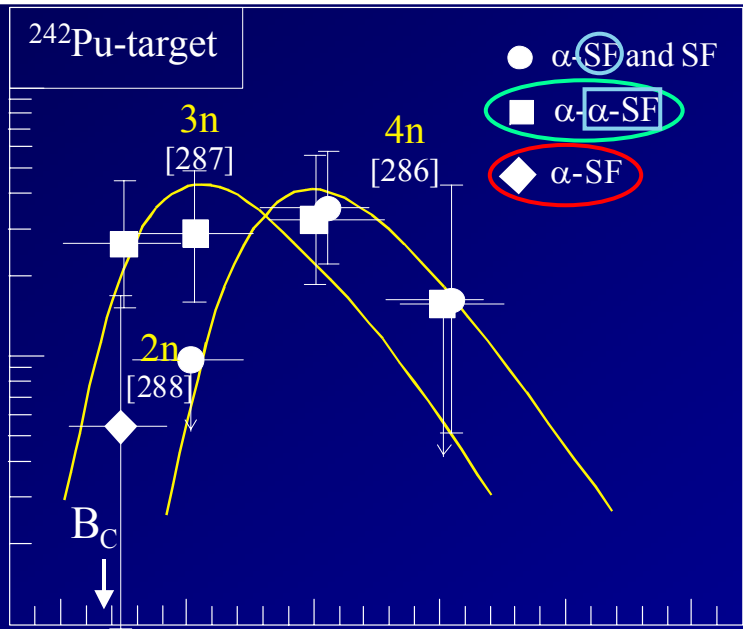
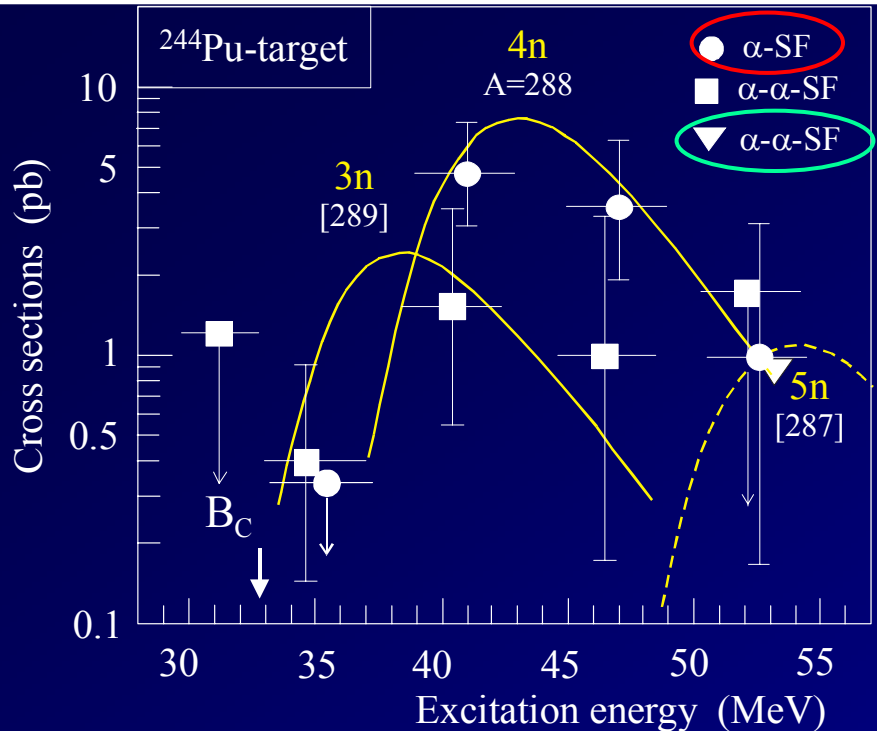


Spectra of alpha particles in the decay chains of isotopes with $Z = 116$



Calculated and Experimental $\text{Log}T_\alpha$ vs Q_α Relationship
for even-even Isotopes with $Z = 98 - 104$





Isotope charge (Z) and mass (A) identifications obtained by the measurements of neutron evaporation cross sections vs. excitation energy of compound nucleus

Z-even nuclei

For Z-odd nuclei
hindrance factor:

for SF- decay ≥ 1000

for α - decay ≤ 10

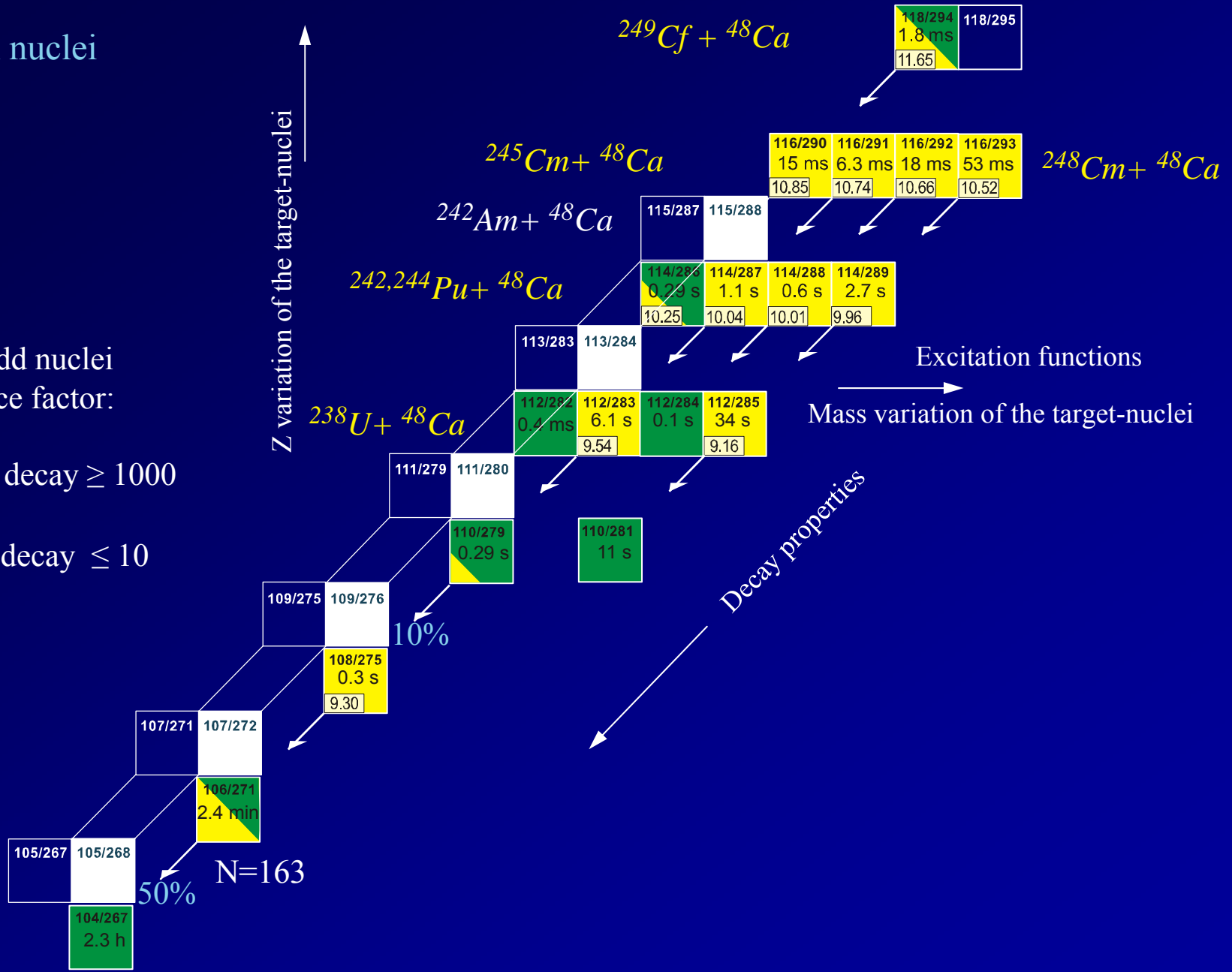
N=162

N=163

Z variation of the target-nuclei

Excitation functions
Mass variation of the target-nuclei

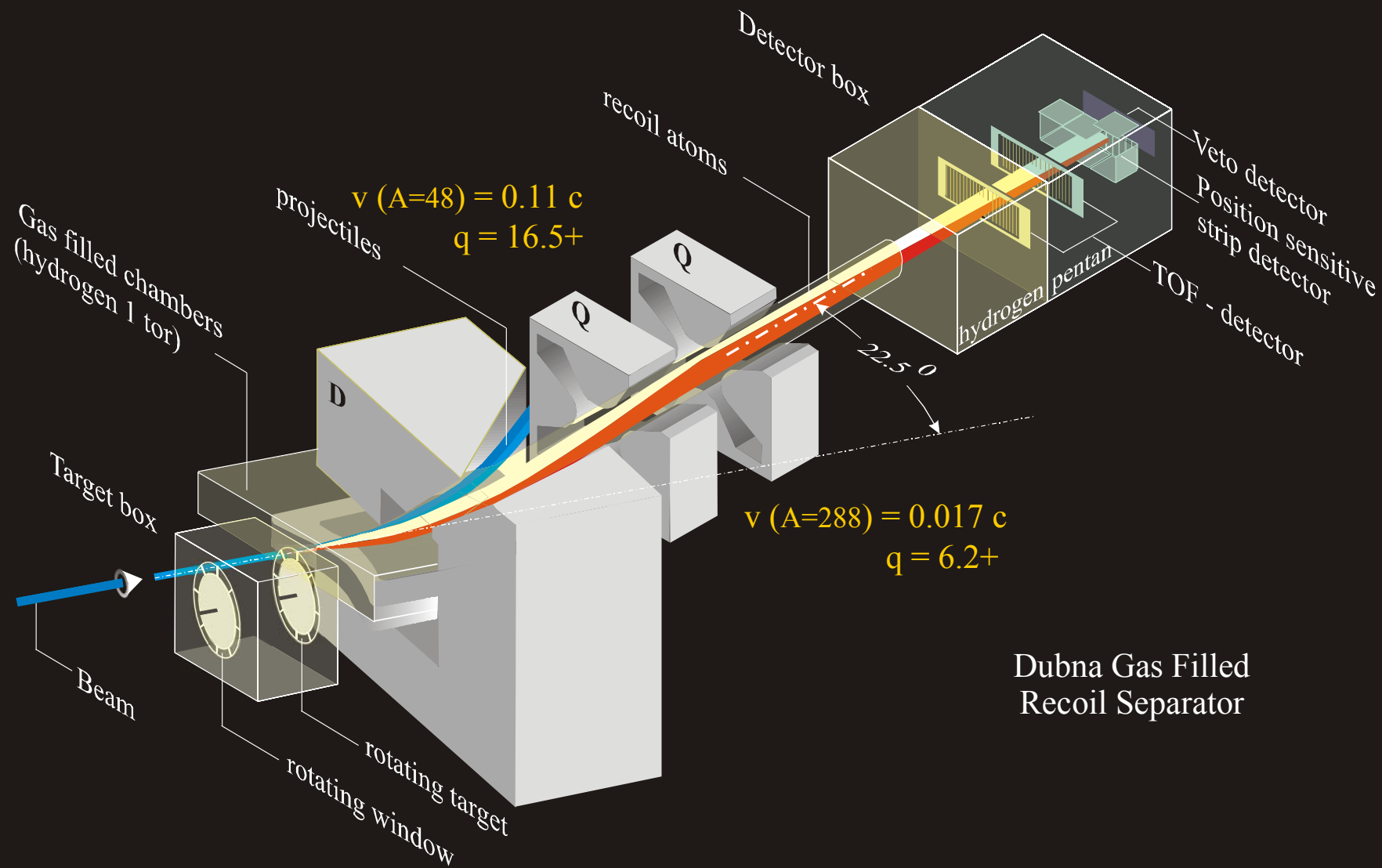
Decay properties



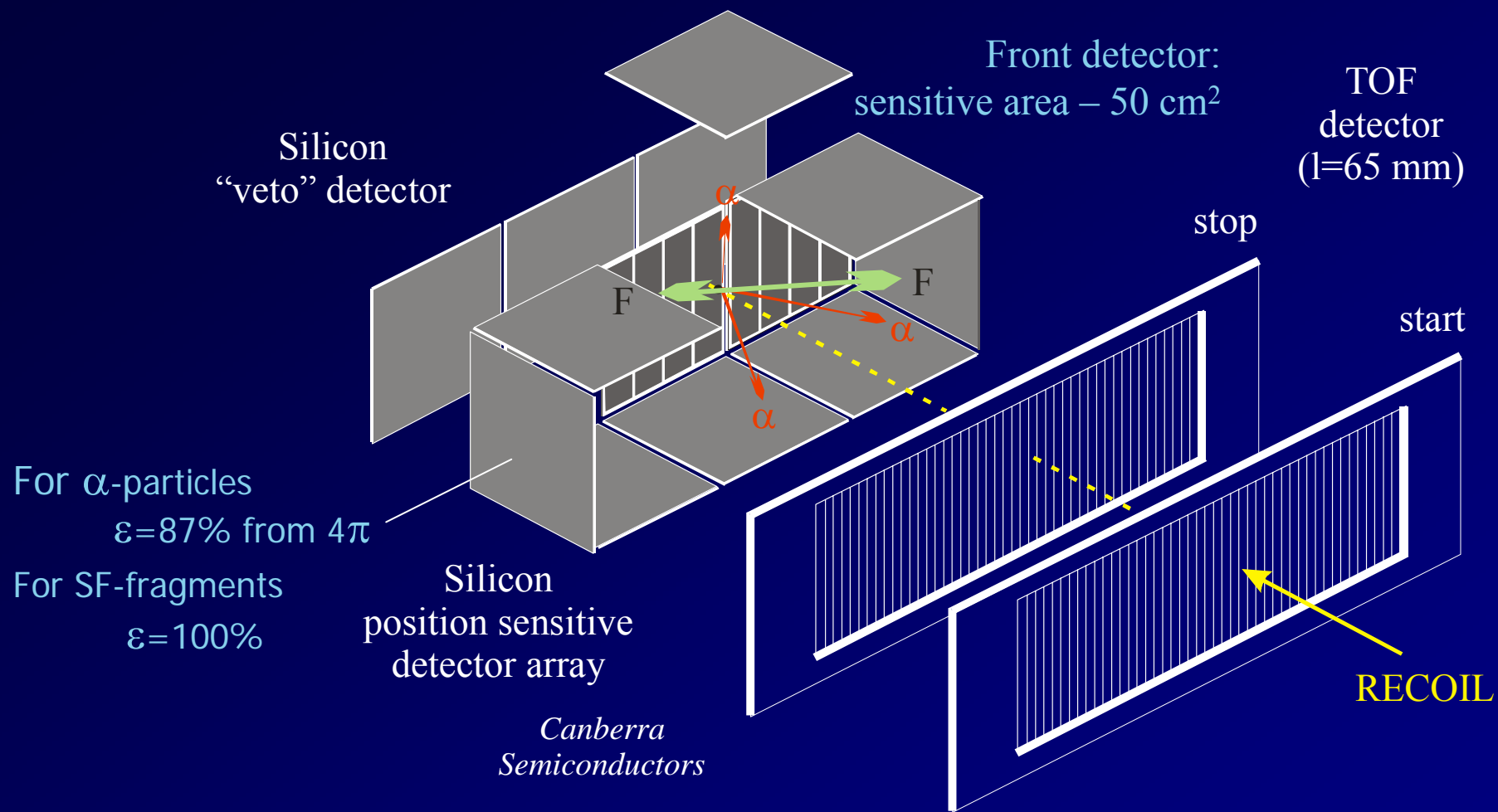
Synthesis of Element 115

in the Reaction:





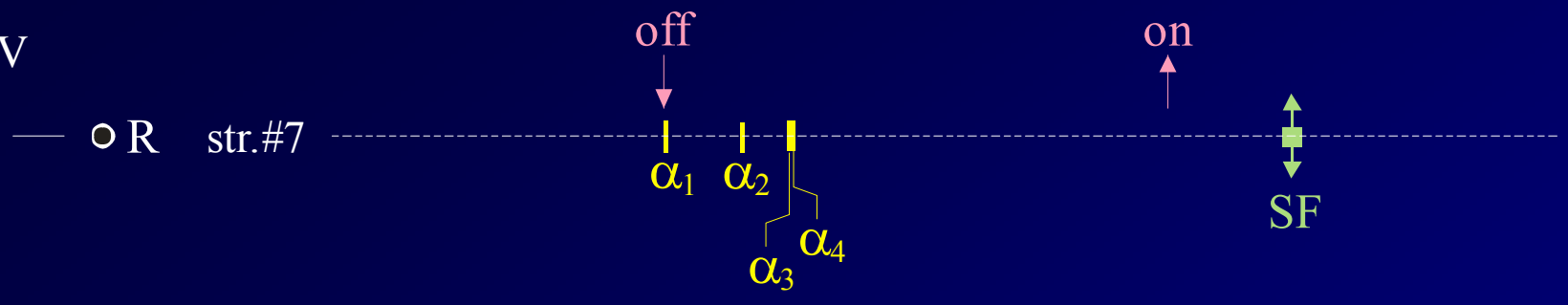
Dubna Gas Filled Recoil Separator



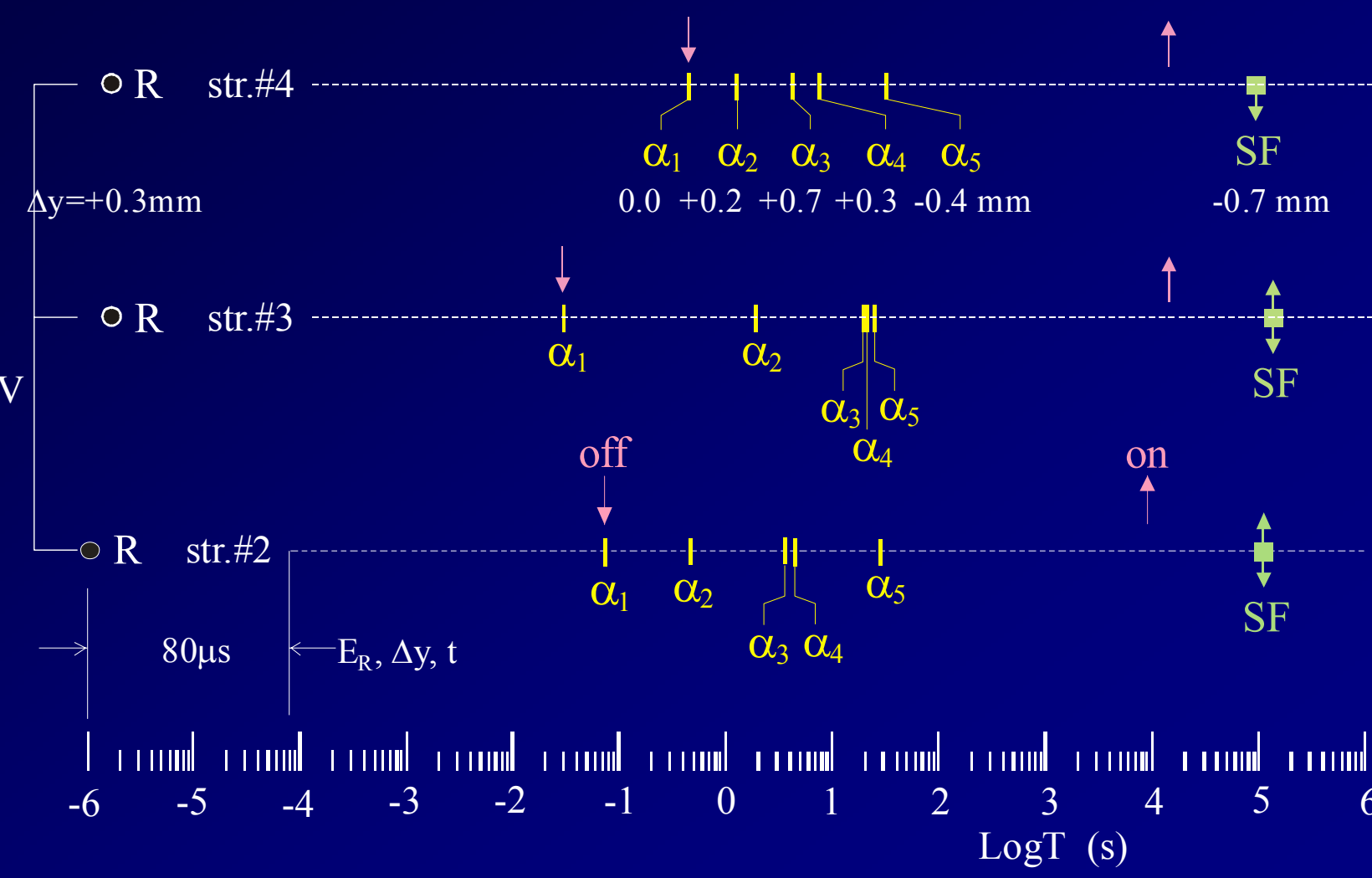
DETECTION SYSTEM
OF THE DUBNA GAS FILLED
RECOIL SEPARATOR

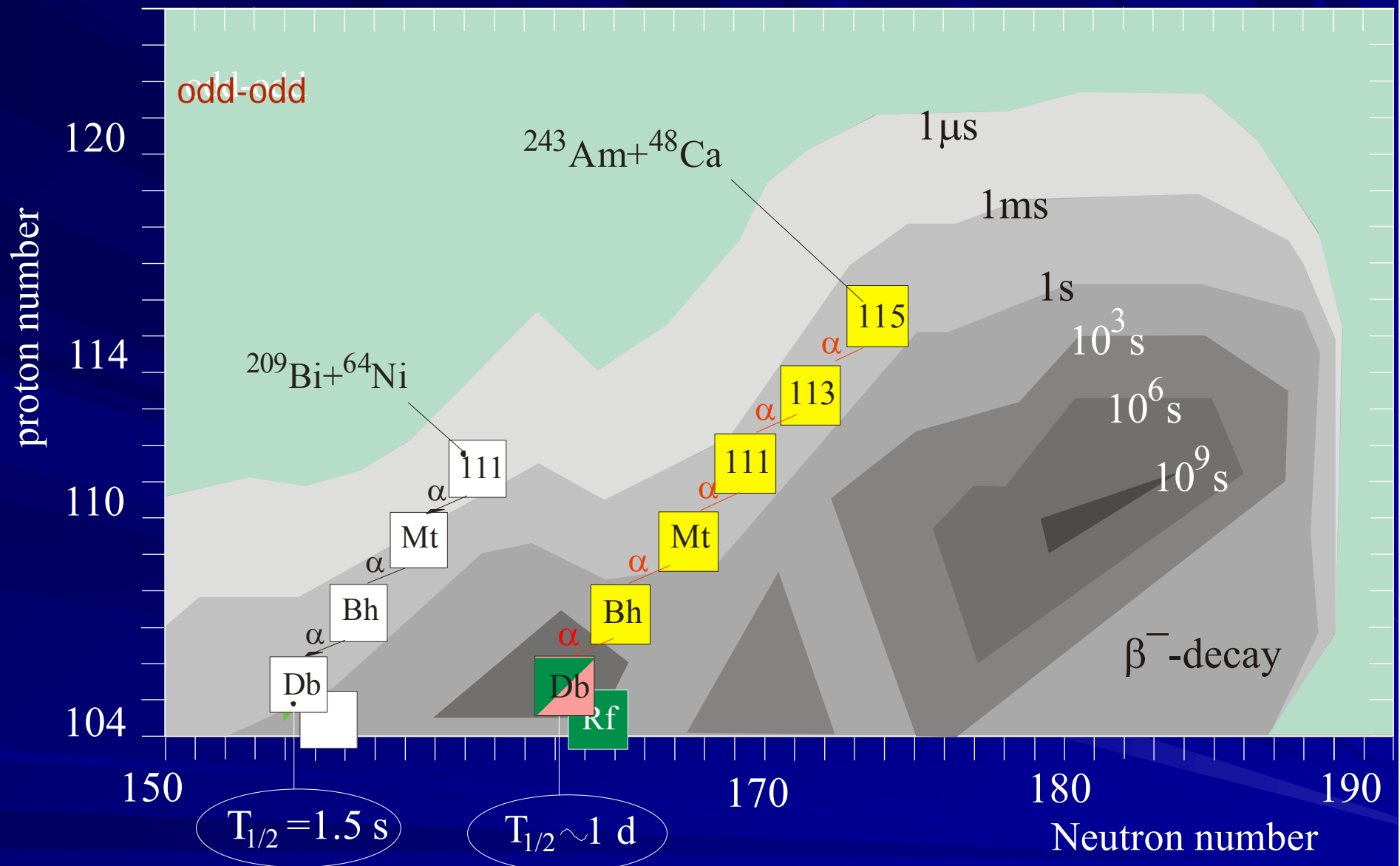
Low pressure multiwire
proportional chamber
pentane - 1.5 Torr

$E_L = 253 \text{ MeV}$



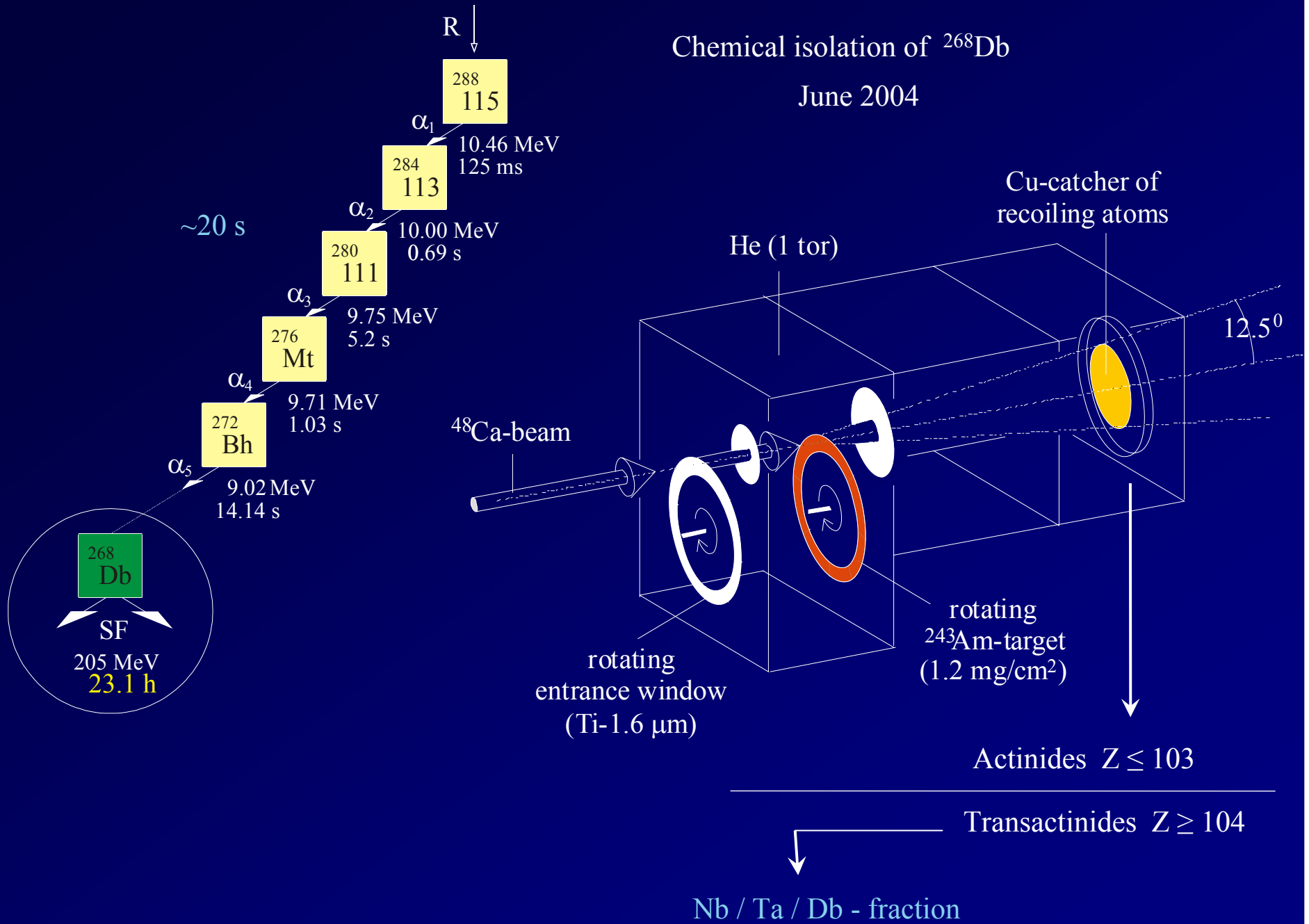
$E_L = 248 \text{ MeV}$



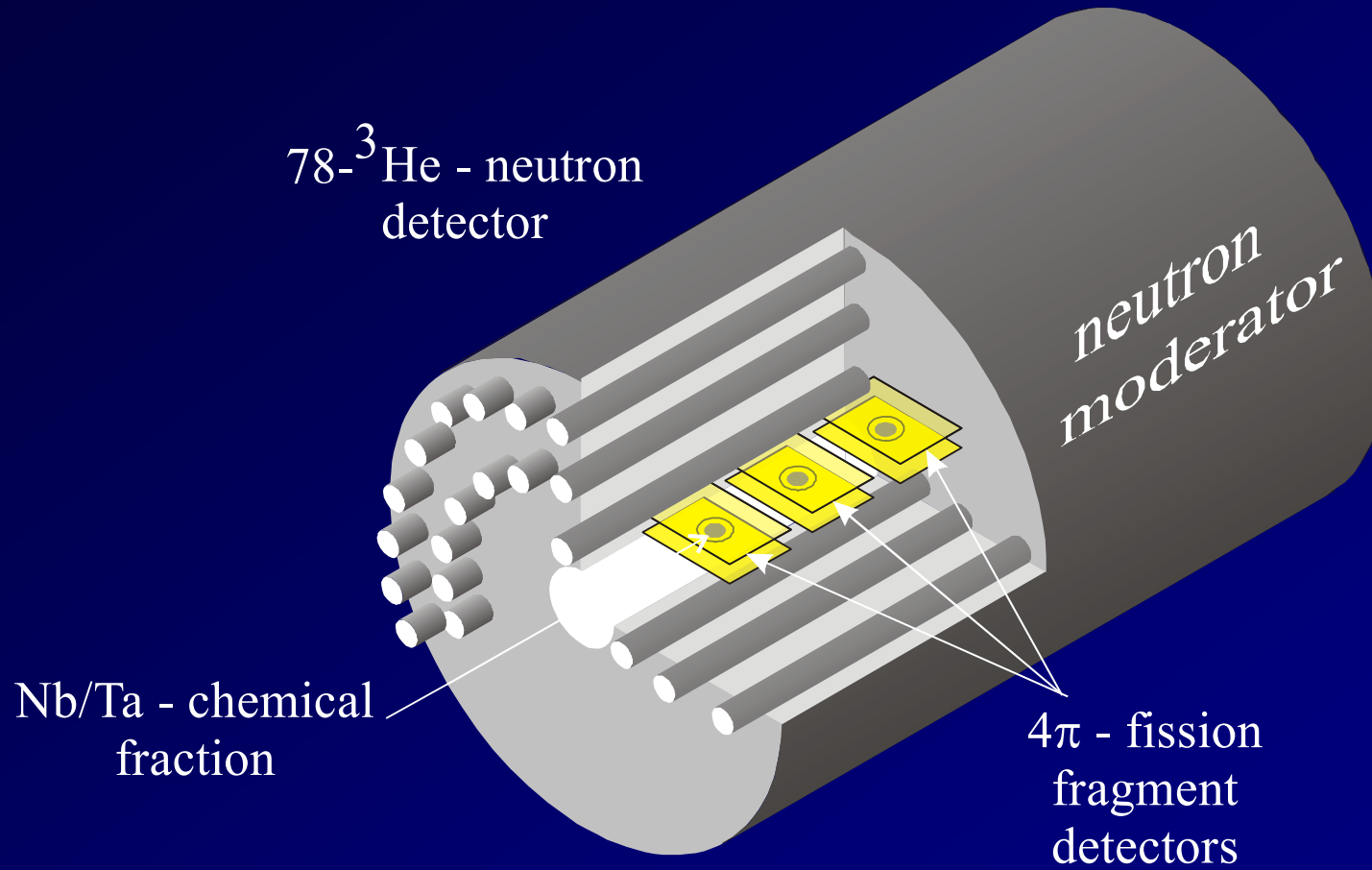


Chemical isolation of ^{268}Db

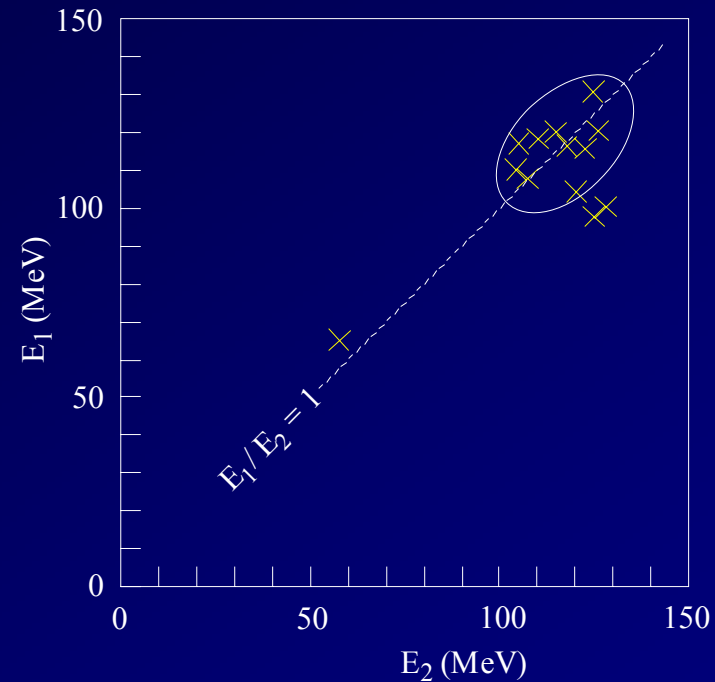
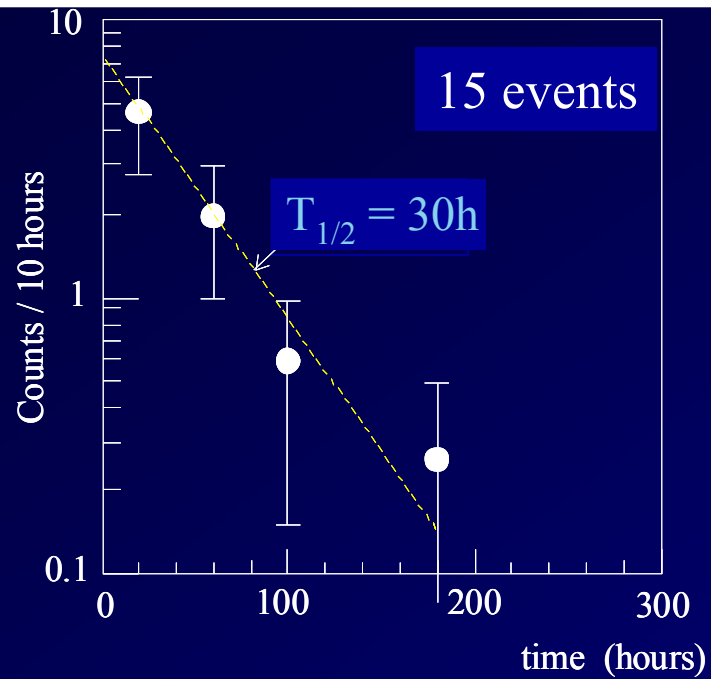
June 2004



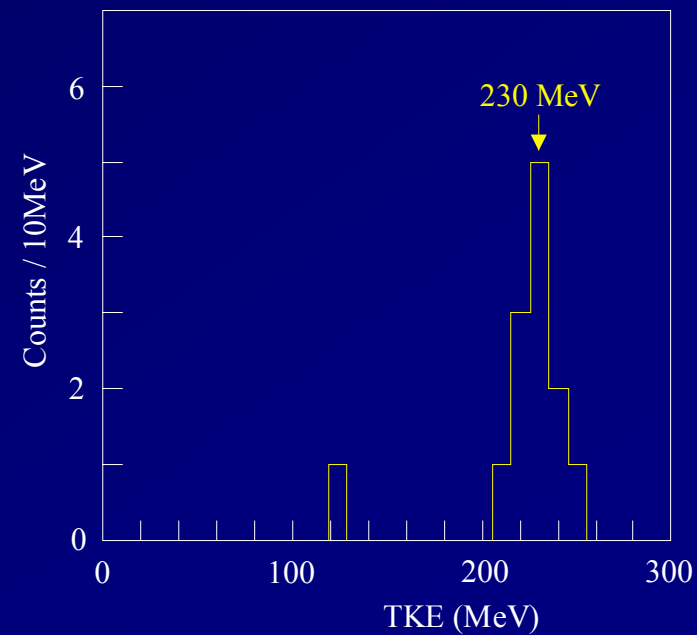
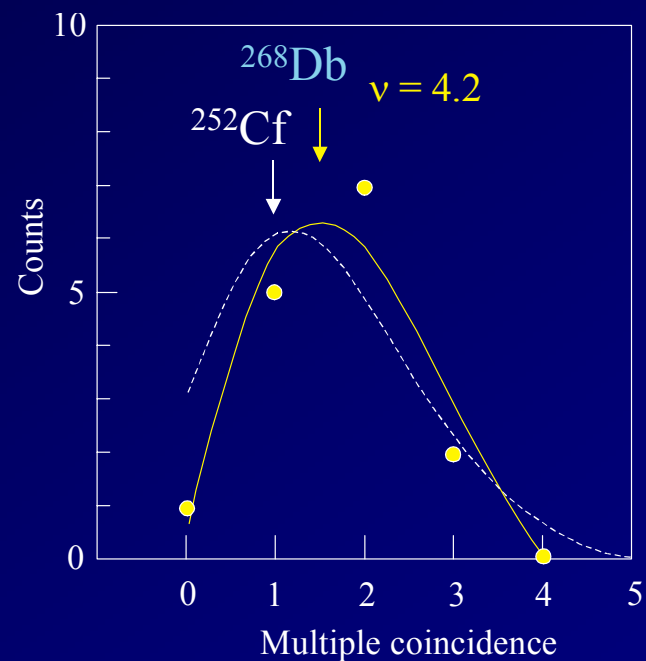
SF of $Z=105$ from the Nb/Ta
chemical fraction

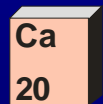


Spontaneous fission
half-life of ^{268}Db
($N=163$)

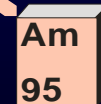
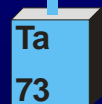


$Q_F \sim 280\text{ MeV}$





Chemical isolation



Gas filled recoil separator

$\leftarrow \alpha$

$\leftarrow \alpha$

$\leftarrow \alpha$

$\leftarrow \alpha$

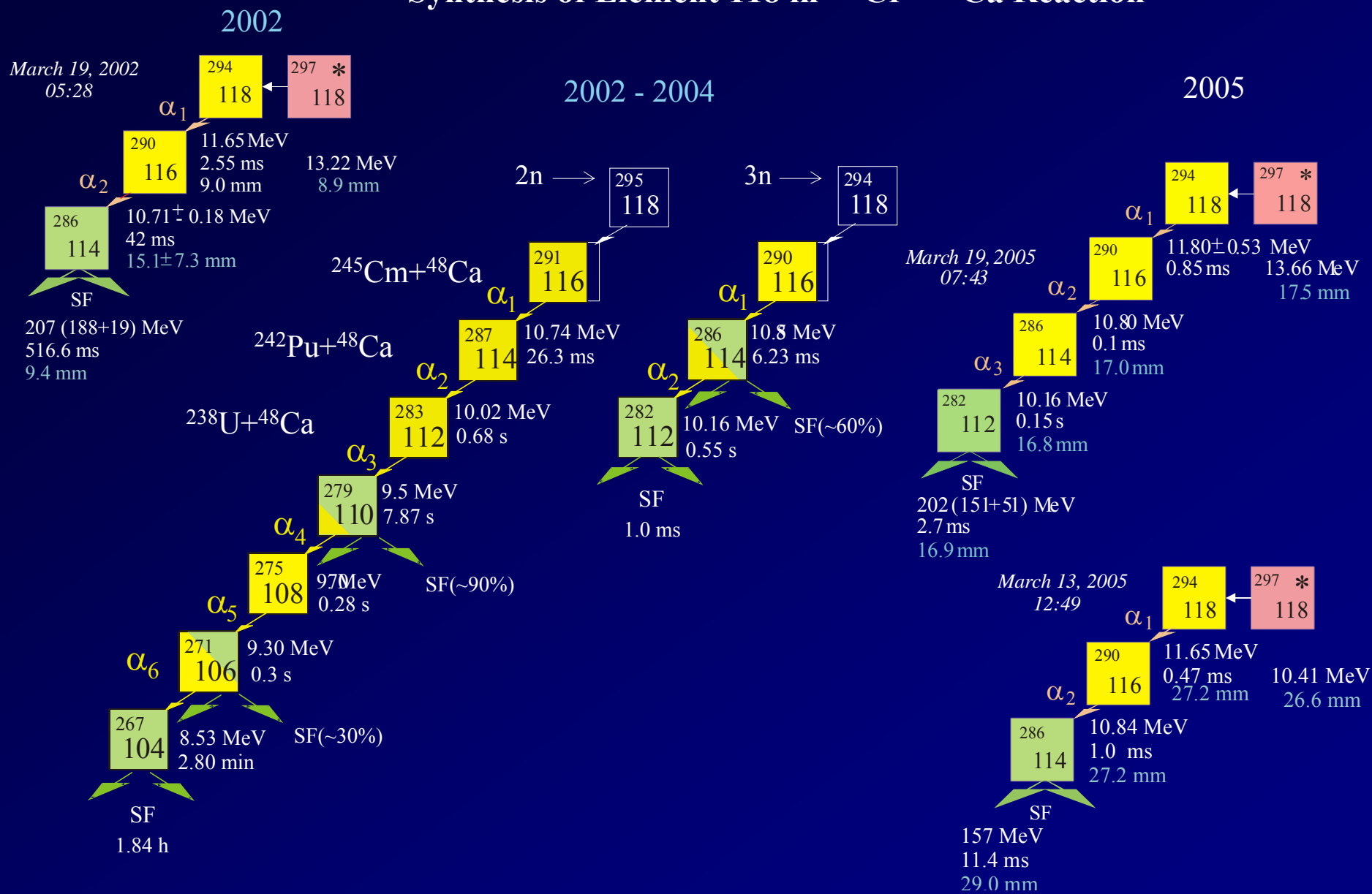
$\leftarrow \alpha$

$\rightarrow n$

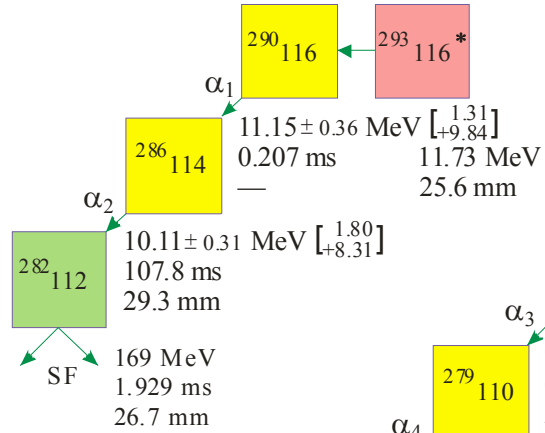
$\uparrow n$

$\downarrow n$

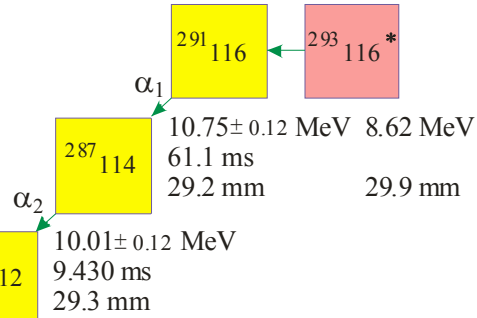
Synthesis of Element 118 in $^{249}\text{Cf} + ^{48}\text{Ca}$ Reaction



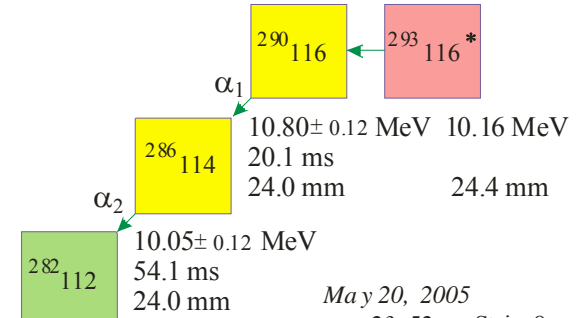
$^{245}\text{Cm} + ^{48}\text{Ca}$ May 09, 2005
 $E^* \sim 38 \text{ MeV}$ 08:51 Strip 4



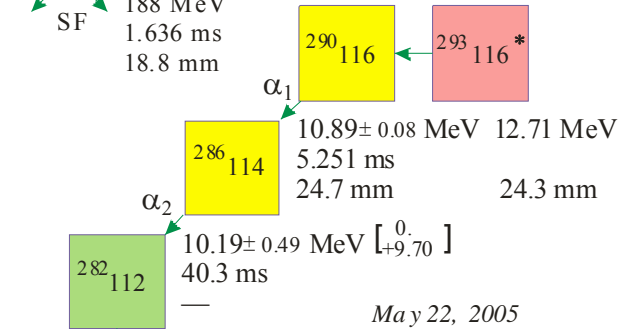
May 14, 2005
 15:25 Strip 5



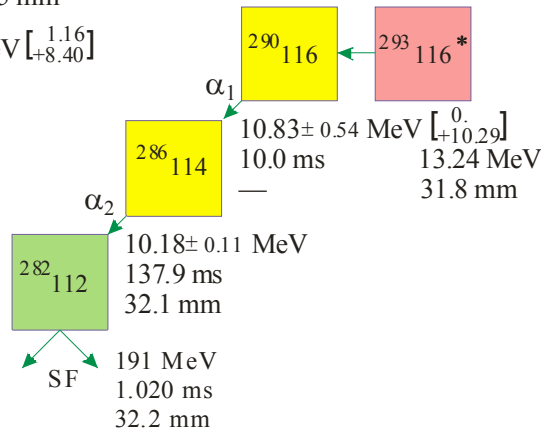
May 17, 2005
 04:27 Strip 5



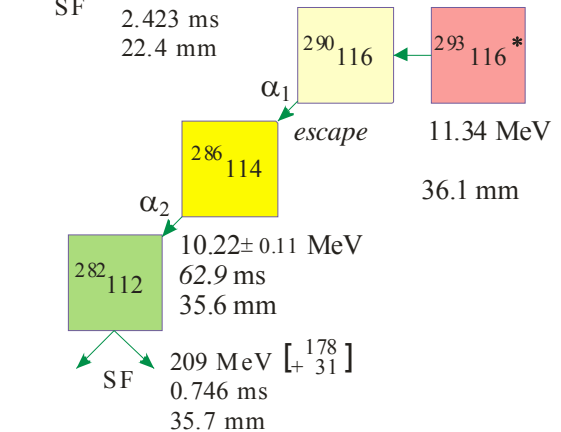
May 20, 2005
 23:52 Strip 8



May 17, 2005
 13:13 Strip 4



May 22, 2005
 18:53 Strip 1



11.15 ± 0.36 MeV [$^{1.31}_{+9.84}$]
 0.207 ms
 11.73 MeV
 25.6 mm

10.11 ± 0.31 MeV [$^{1.80}_{+8.31}$]
 107.8 ms
 29.3 mm

169 MeV
 1.929 ms
 26.7 mm

9.57 ± 0.12 MeV
 0.0346 s
 29.3 mm

9.70 ± 0.12 MeV
 0.7070 s
 29.5 mm

9.56 ± 0.34 MeV [$^{1.16}_{+8.40}$]
 0.3679 s

8.84 ± 0.36 MeV [$^{3.43}_{+5.41}$]
 94.1035 s
 31.1 mm

240 MeV [$^{193}_{+47}$]
 20.99755 s
 29.3 mm

10.75 ± 0.12 MeV 8.62 MeV
 61.1 ms
 29.2 mm 29.9 mm

10.01 ± 0.12 MeV
 9.430 ms
 29.3 mm

10.80 ± 0.12 MeV 10.16 MeV
 20.1 ms
 24.0 mm 24.4 mm

10.05 ± 0.12 MeV
 54.1 ms
 24.0 mm

188 MeV
 1.636 ms
 18.8 mm

10.89 ± 0.08 MeV 12.71 MeV
 5.251 ms
 24.7 mm 24.3 mm

10.19 ± 0.49 MeV [$^{0.}_{+9.70}$]
 40.3 ms

194 MeV
 2.423 ms
 22.4 mm

10.83 ± 0.54 MeV [$^{0.}_{+10.29}$]
 10.0 ms
 13.24 MeV
 31.8 mm

10.18 ± 0.11 MeV
 137.9 ms
 32.1 mm

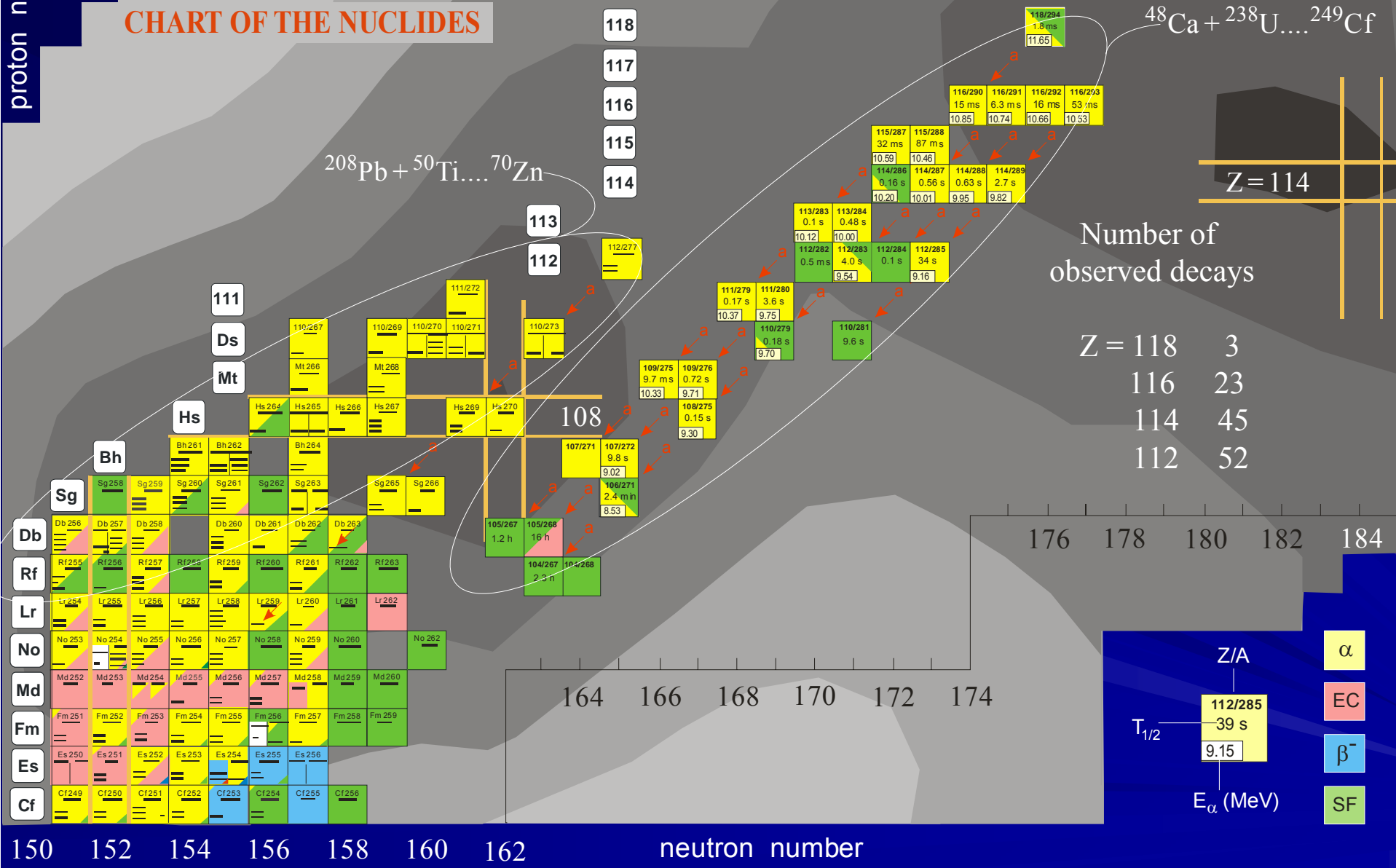
191 MeV
 1.020 ms
 32.2 mm

10.22 ± 0.11 MeV
 62.9 ms
 35.6 mm

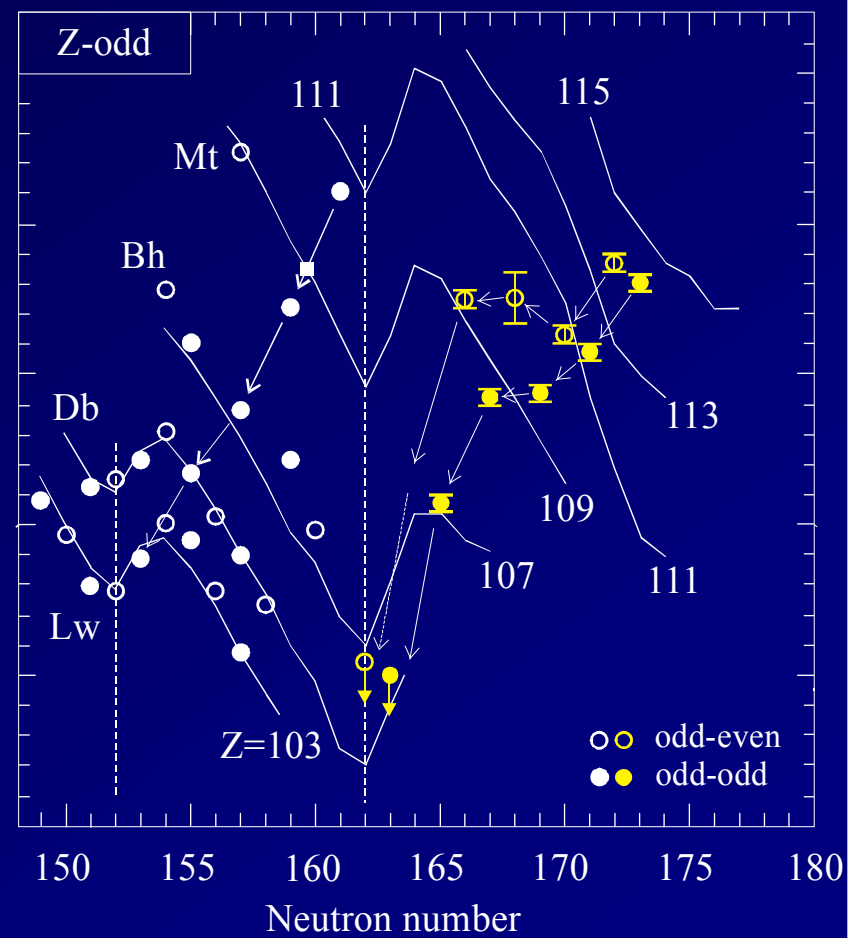
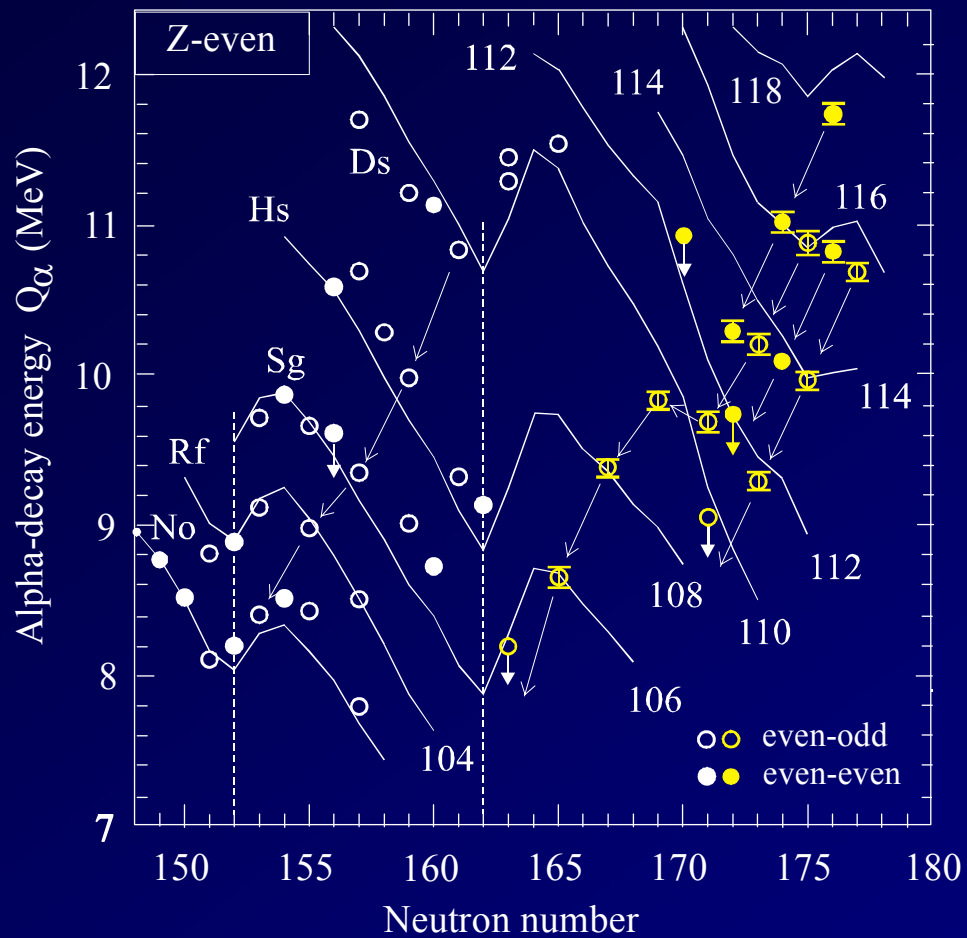
209 MeV [$^{178}_{+31}$]
 0.746 ms
 35.7 mm

proton number

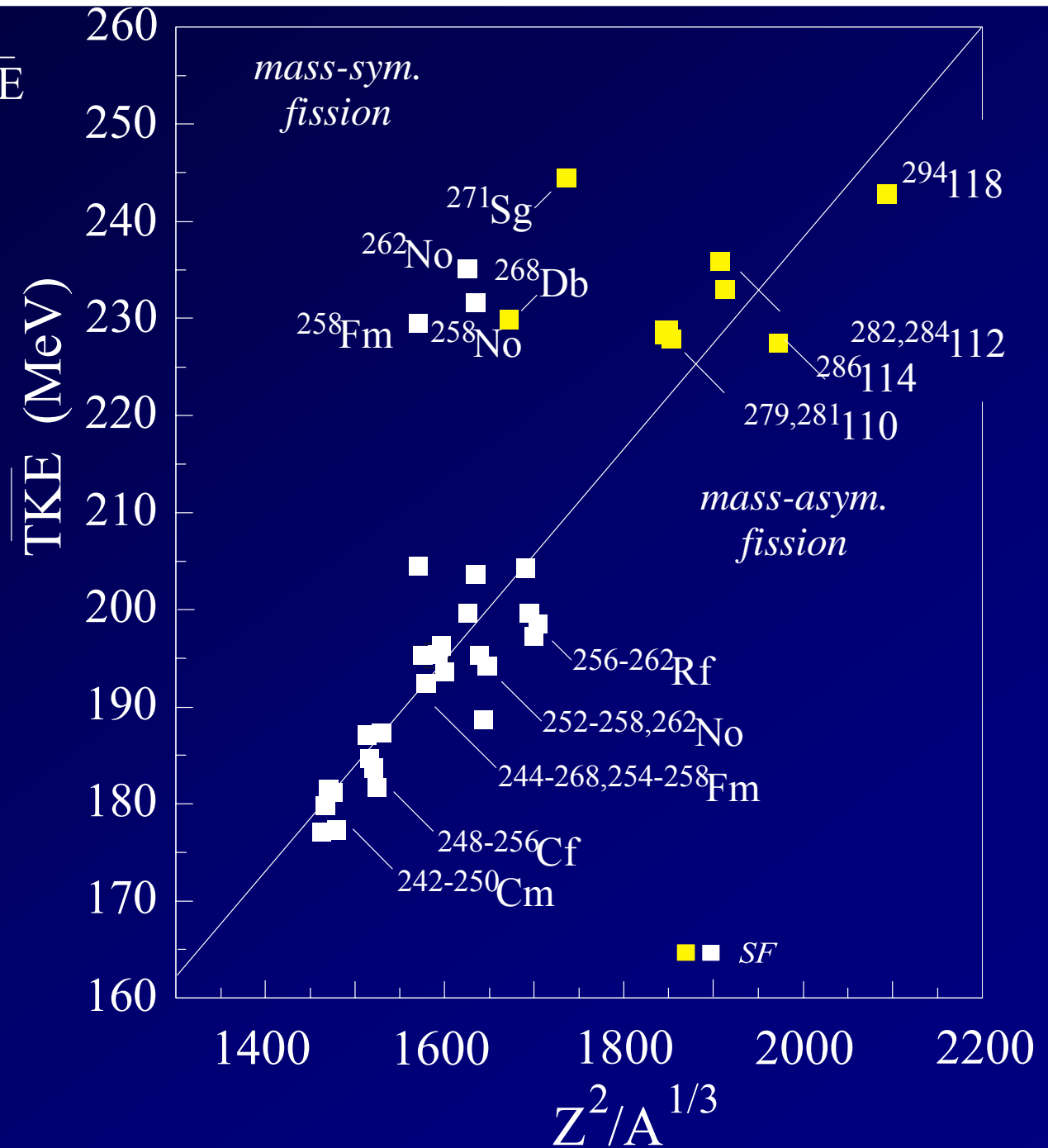
CHART OF THE NUCLIDES



Alpha-decay energy vs. neutron number For the isotopes of elements with $Z > 102$

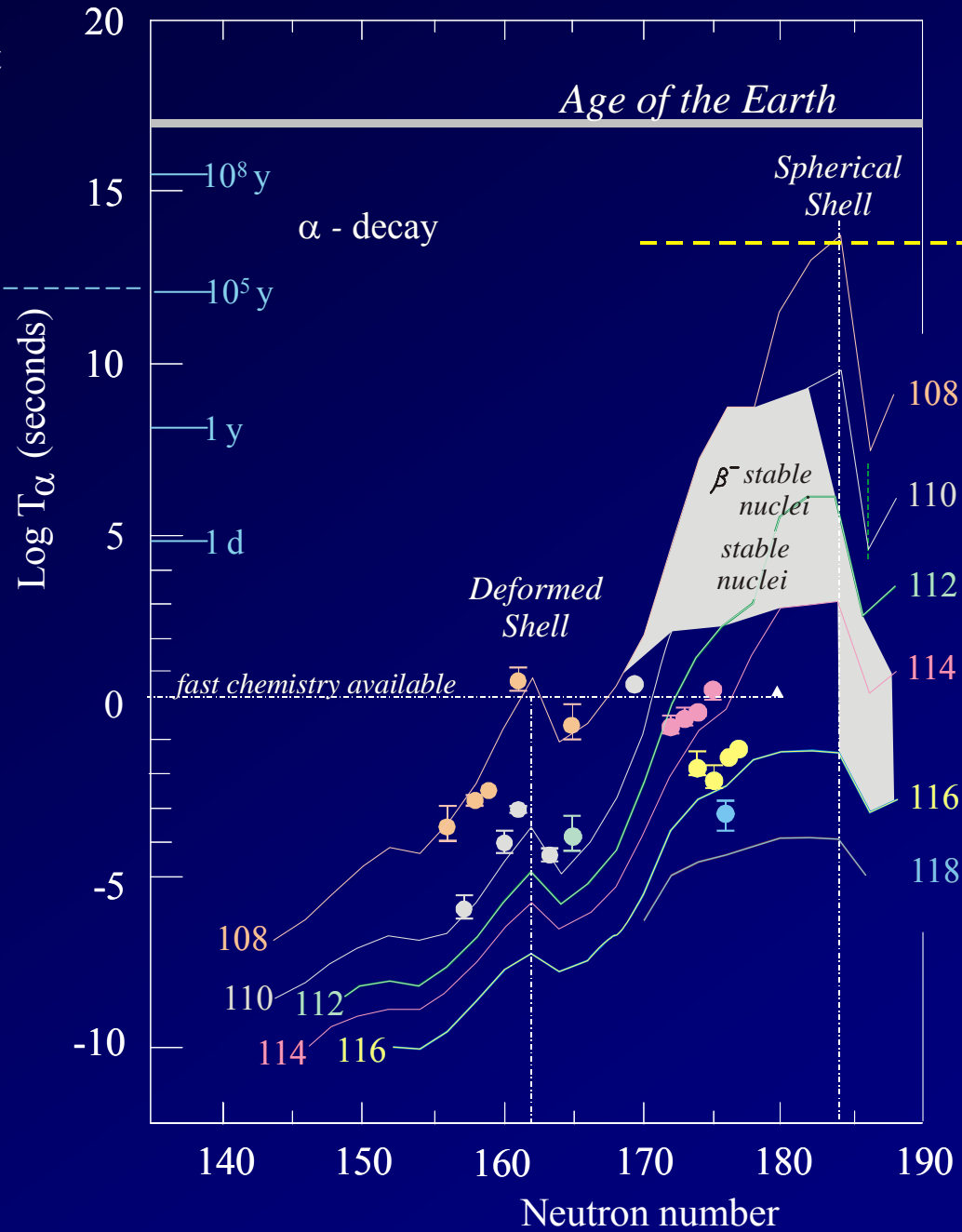


SF-systematic of $\overline{\text{TKE}}$



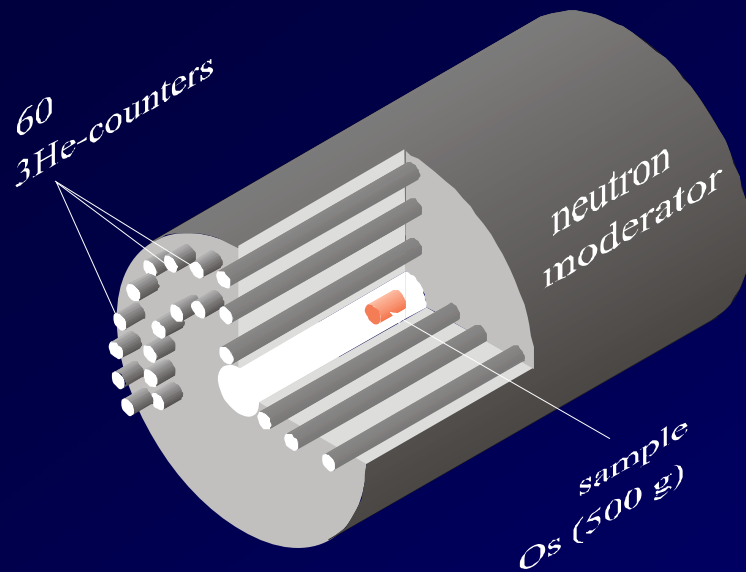
Theory and Experiment

the search for SHE
in Cosmic rays



the experimental
limit of the search
for SHE in Nature

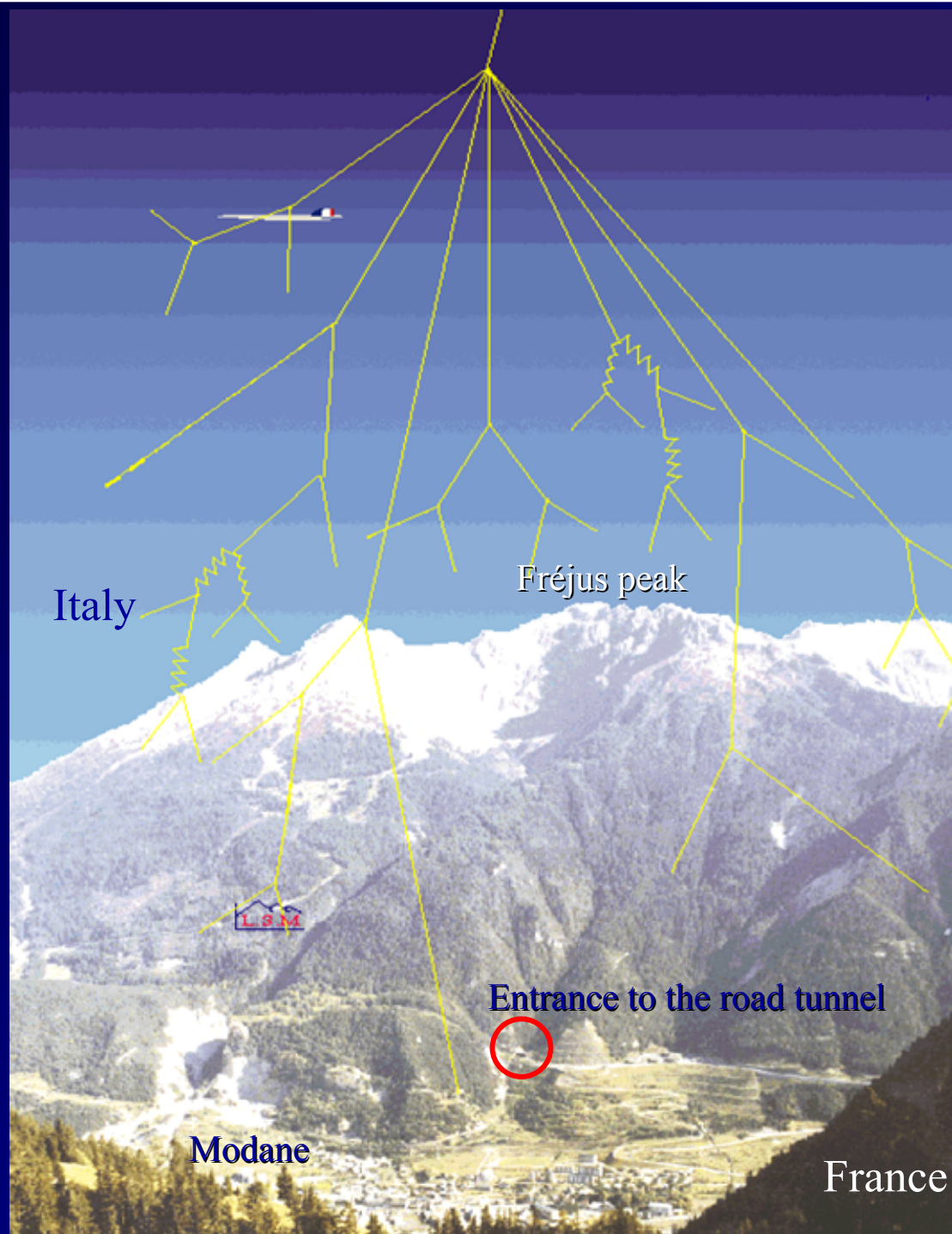
Search for SF of natural Eka Os
by detection of fission neutrons

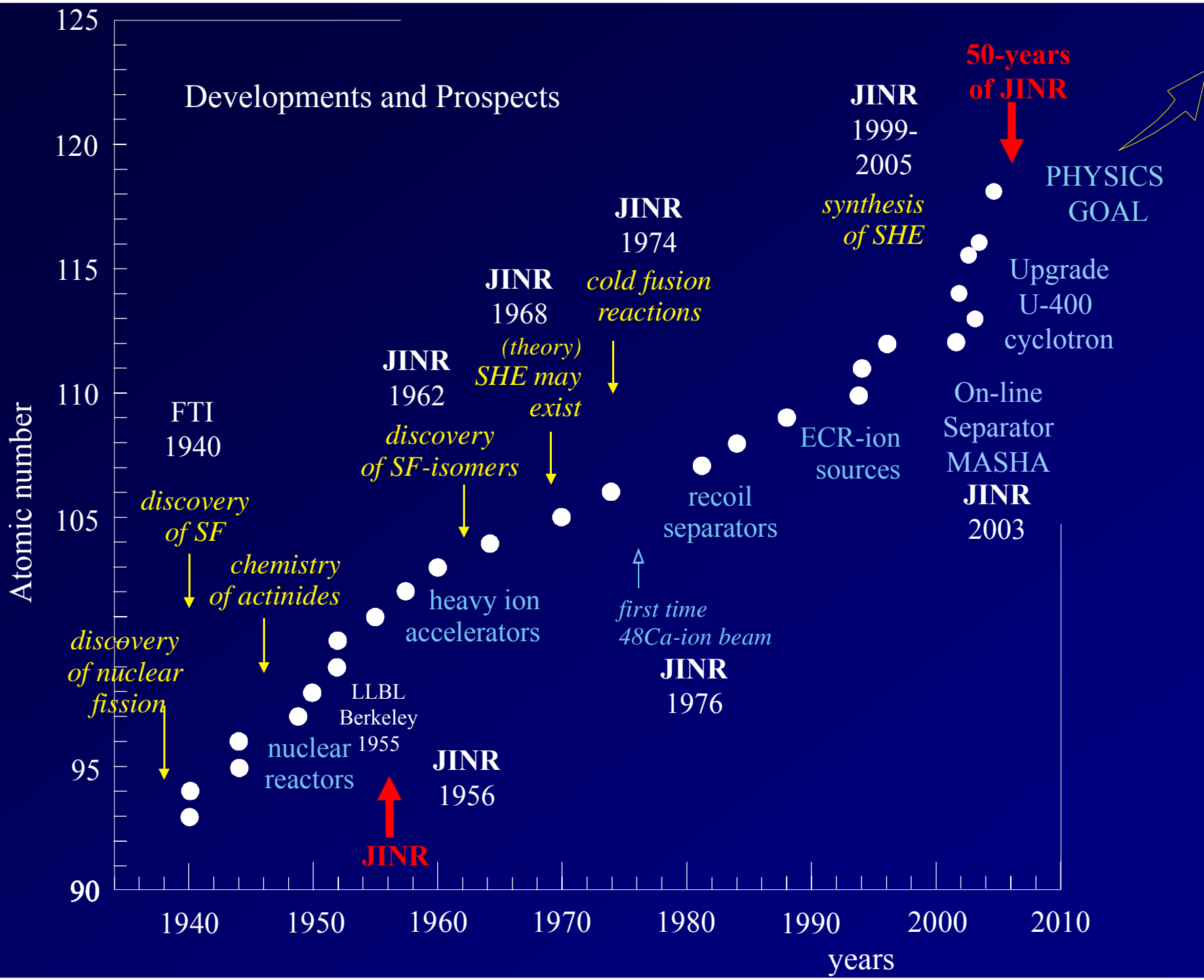


1 SF-event per year ($T_{1/2}=10^9\text{y}$)
corresponds to the concentration:

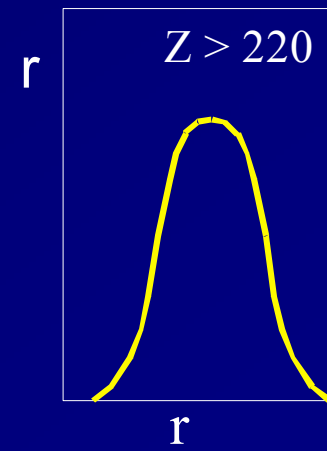
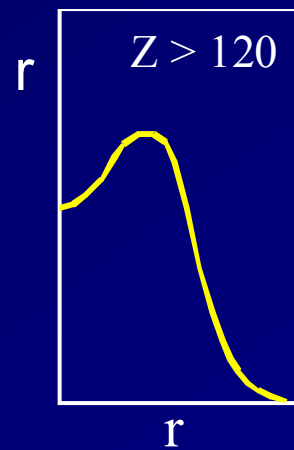
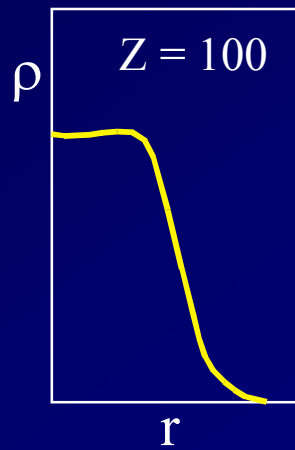
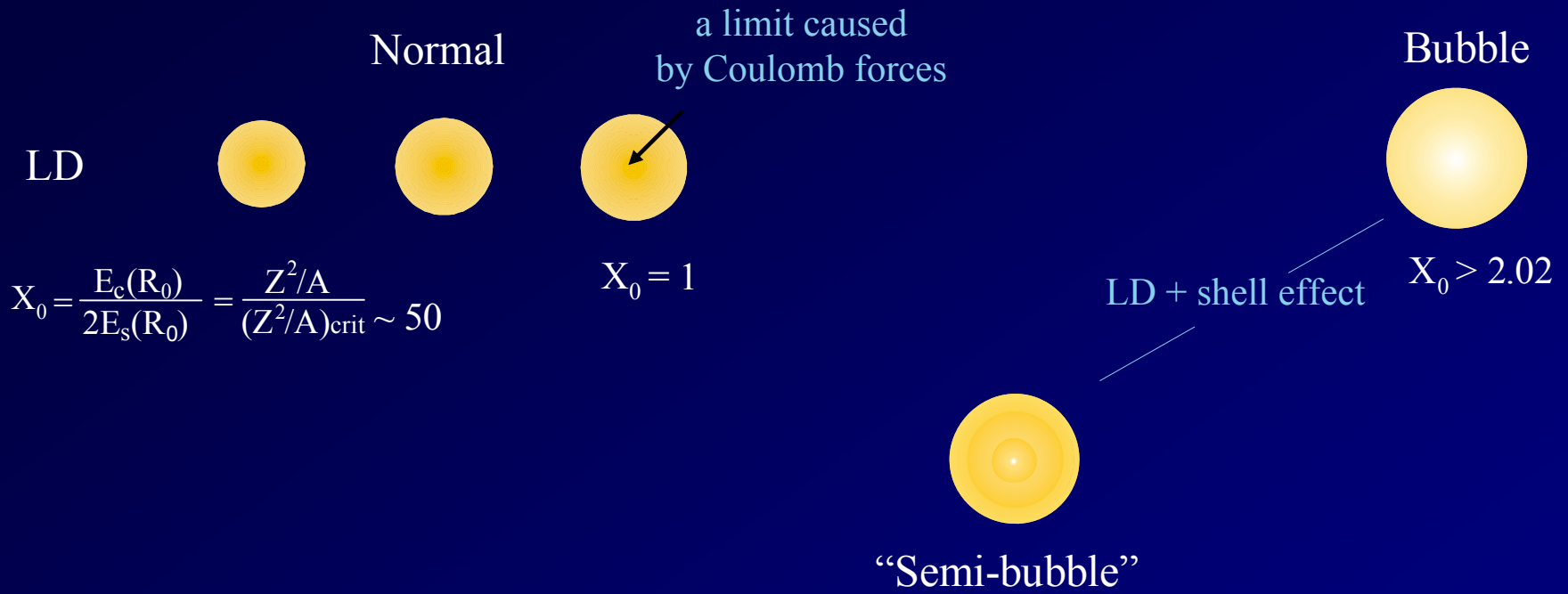
$$\text{EkaOs/Os} = 5 \cdot 10^{-15} \text{g/g}$$

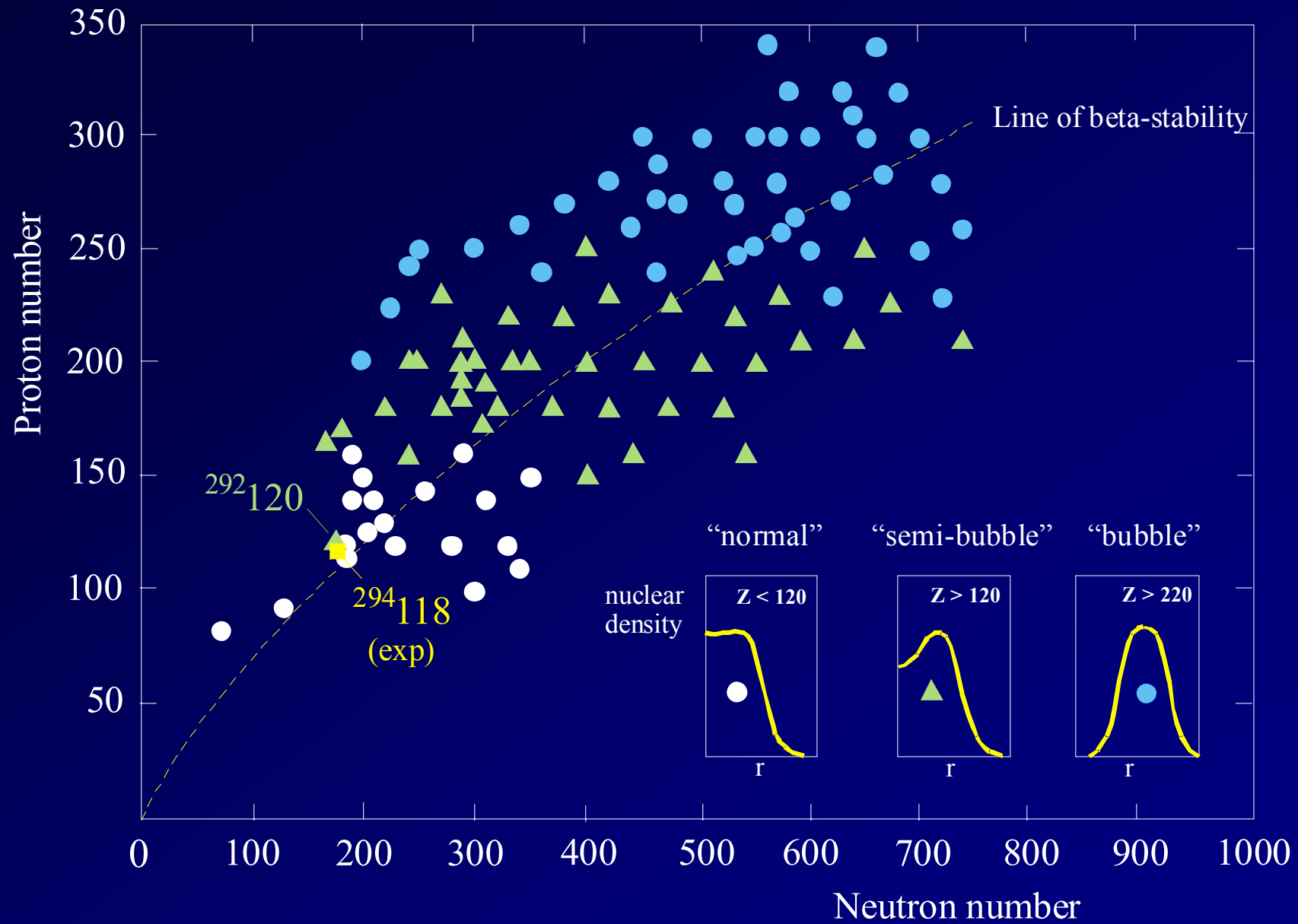
(or 10^{-22}g/g in the terrestrial matter,
or 10^{-16} of U)

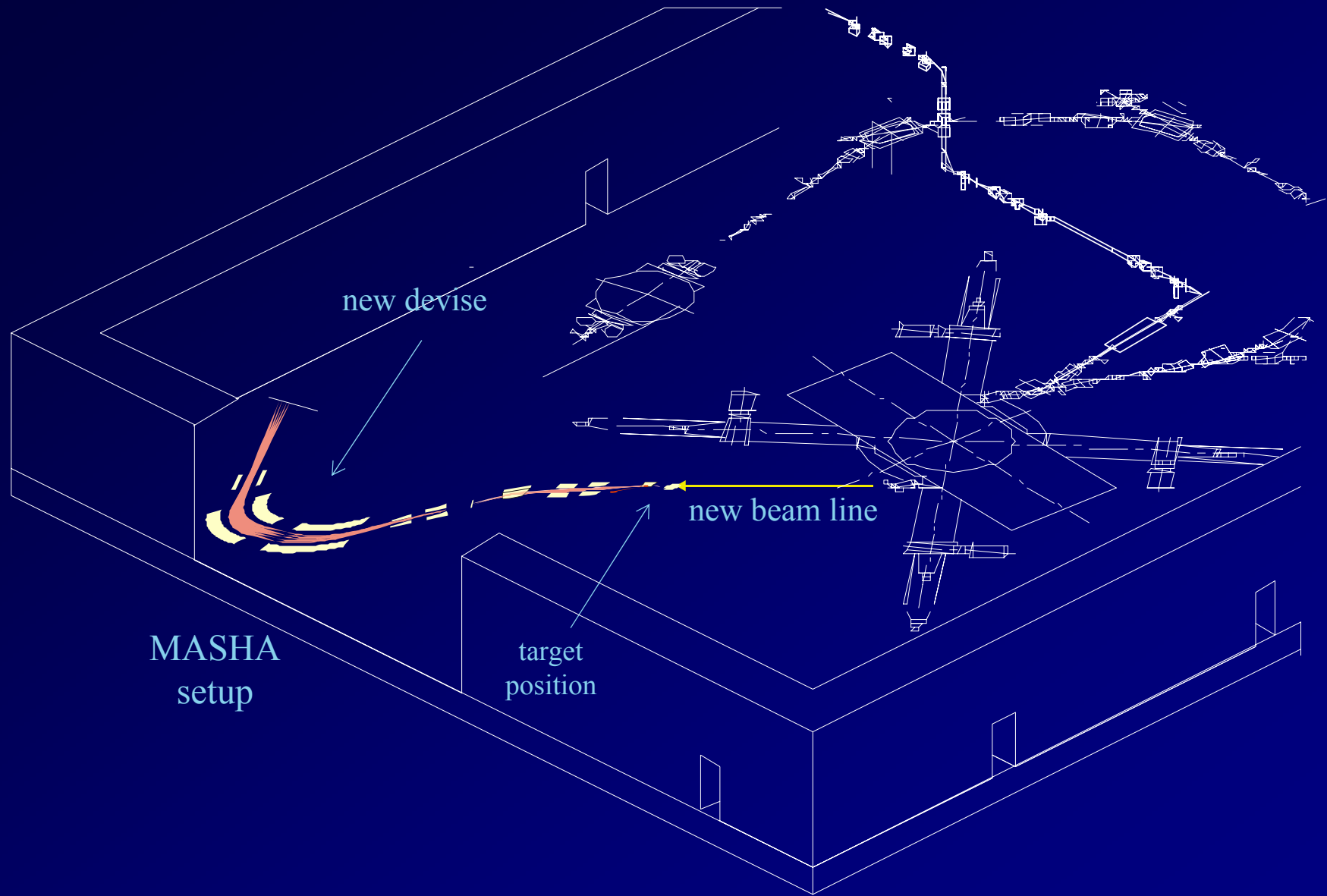




Nuclear Exotica in Superheavy Nuclei







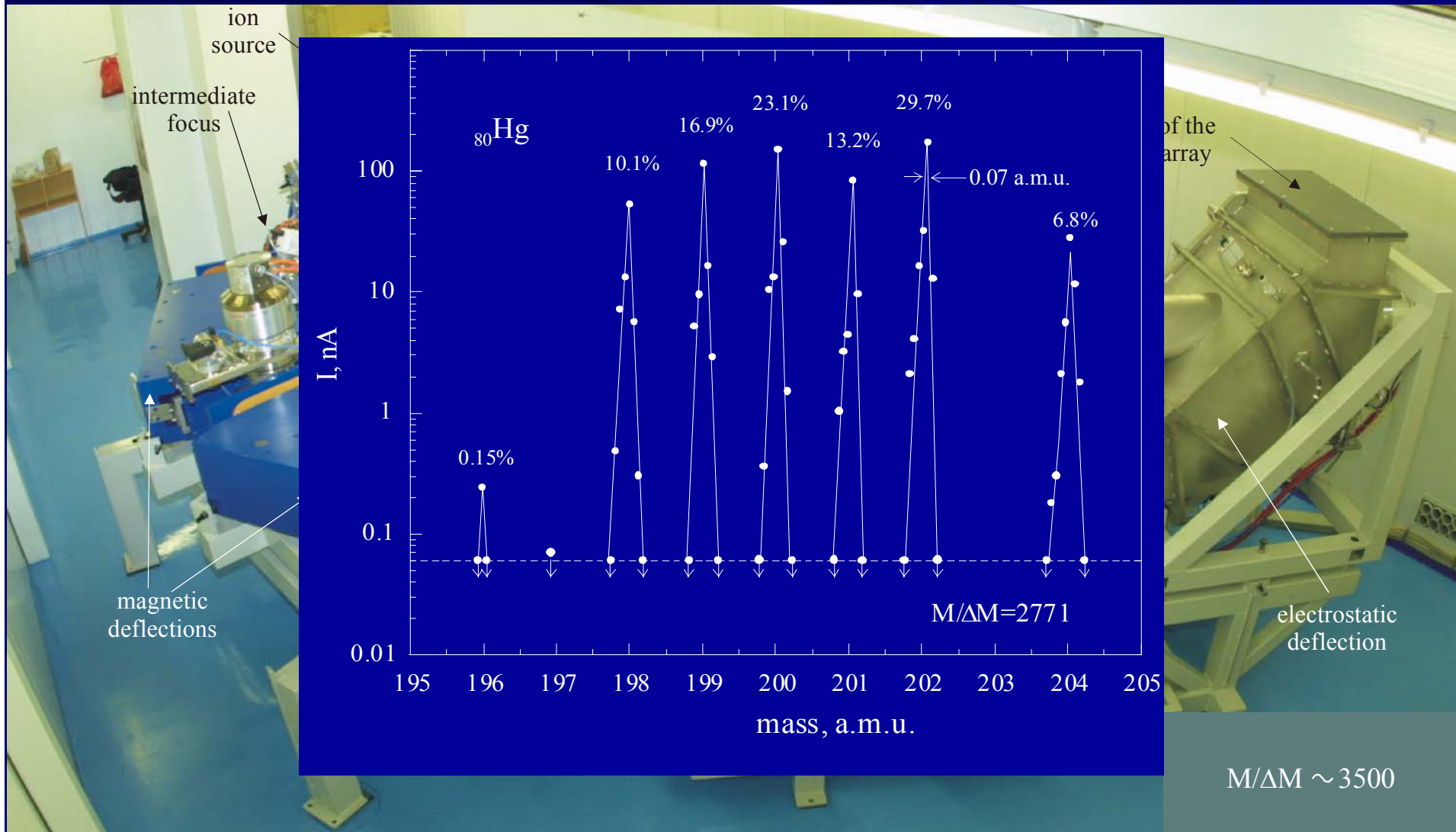
MASHA
setup

new devise

new beam line

target
position

Mass Analyzer of Super Heavy Atoms (MASHA)



Periodic Table of the Elements

1																	18	
1	H 1 Hydrogen											He 2 Helium						
2	Li 3 Lithium	Be 4 Beryllium											B 5 Boron	C 6 Carbon	N 7 Nitrogen	O 8 Oxygen	F 9 Fluorine	Ne 10 Neon
3	Na 11 Sodium	Mg 12 Magnesium											Al 13 Aluminum	Si 14 Silicon	P 15 Phosphorus	S 16 Sulfur	Cl 17 Chlorine	Ar 18 Argon
4	K 19 Potassium	Ca 20 Calcium	Sc 21 Scandium	Ti 22 Titanium	V 23 Vanadium	Cr 24 Chromium	Mn 25 Manganese	Fe 26 Iron	Co 27 Cobalt	Ni 28 Nickel	Cu 29 Copper	Zn 30 Zinc	Ga 31 Gallium	Ge 32 Germanium	As 33 Arsenic	Se 34 Selenium	Br 35 Bromine	Kr 36 Krypton
5	Rb 37 Rubidium	Sr 38 Strontium	Y 39 Yttrium	Zr 40 Zirconium	Nb 41 Niobium	Mo 42 Molybdenum	Tc 43 Technetium	Ru 44 Ruthenium	Rh 45 Rhodium	Pd 46 Palladium	Ag 47 Silver	Cd 48 Cadmium	In 49 Indium	Sn 50 Tin	Sb 51 Antimony	Te 52 Tellurium	I 53 Iodine	Xe 54 Xenon
6	Cs 55 Cesium	Ba 56 Barium	La 57 Lanthanum	Hf 72	Ta 73	W 74	Re 75	Os 76	Ir 77 Iridium	Pt 78 Platinum	Au 79 Gold	Hg 80 Mercury	Tl 81 Thallium	Pb 82 Lead	Bi 83 Bismuth	Po 84 Polonium	At 85 Astatine	Rn 86 Radon
7	Fr 87 Francium	Ra 88 Radium	Ac 89 Actinium					Hs 108 Hassium										
												112	113	114	115	116	117	118

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Elements with $Z \geq 120$

Actinides

Ac 89 Actinium	Th 90 Thorium	Pa 91 Protactinium	U 92 Uranium	Np 93 Neptunium	Pu 94 Plutonium	Am 95 Americium
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Chemical properties (relativistic effect)

Astrophysics (search for SHE in cosmic rays)

Nucleosynthesis (test of the r - s process)

Atomic physics (structure of SH-atoms)



**Flerov Laboratory of Nuclear Reactions
of JINR**

...in February

Thanks for your attention